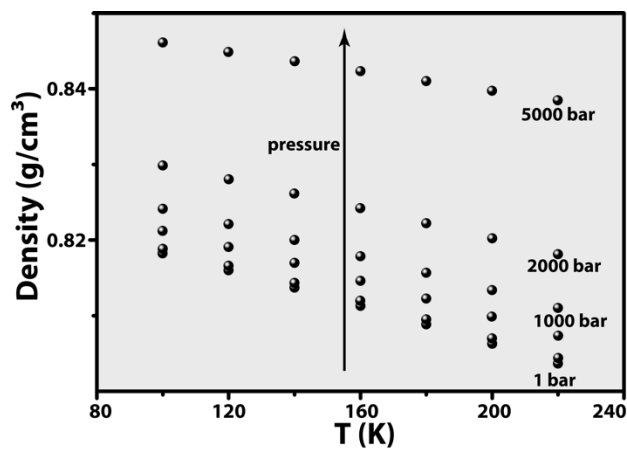


*Low-temperature Thermodynamic Study of the  
Metastable Empty Clathrate Hydrates using  
Molecular Simulations*

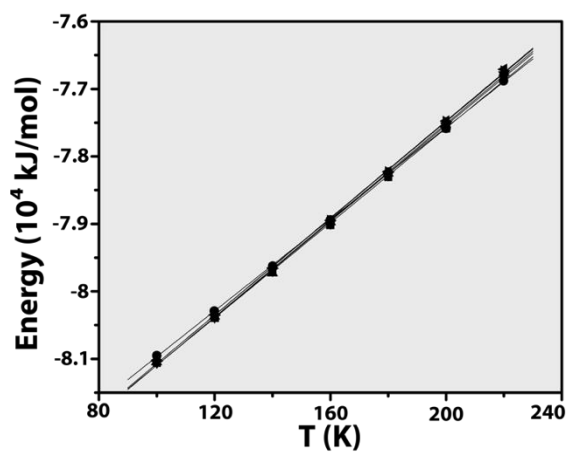
*Fernando J.A.L. Cruz<sup>1,\*</sup>, Saman Alavi<sup>2</sup> and José P.B. Mota<sup>1</sup>*

<sup>1</sup>LAQV-REQUIMTE, Department of Chemistry, Faculdade de Ciências e Tecnologia,  
Universidade NOVA de Lisboa, 2829-516 Caparica, Portugal. <sup>2</sup>National Research Council of  
Canada, Ottawa, Ontario, K1A 0R6, Canada. <sup>3</sup>Department of Chemistry and Biomolecular  
Sciences, University of Ottawa, Ottawa, Ontario, K1N 6N5, Canada

\*[fj.cruz@fct.unl.pt](mailto:fj.cruz@fct.unl.pt)



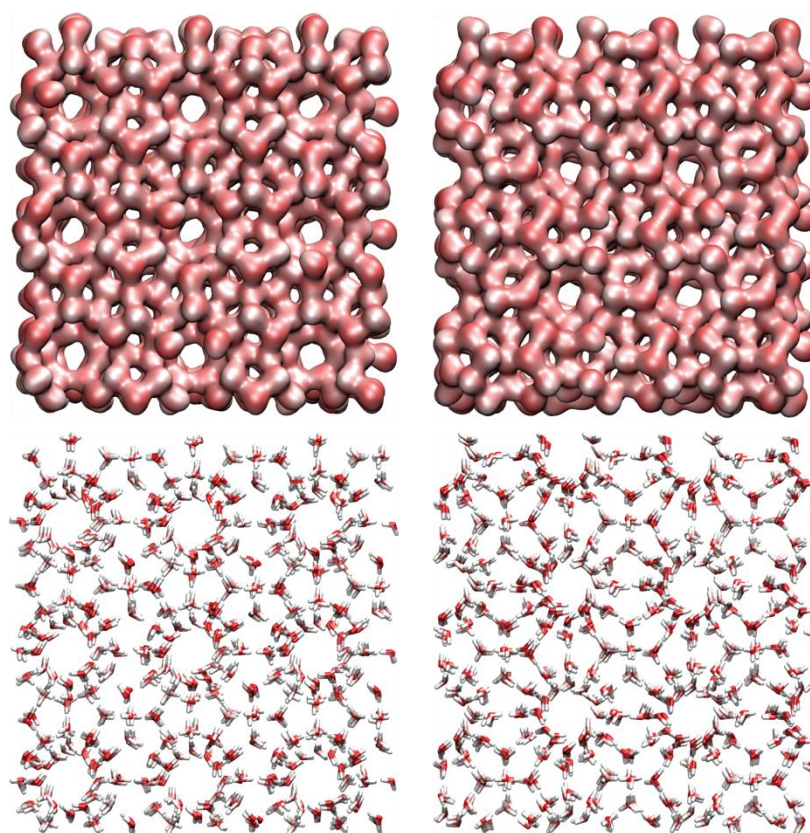
**Figure S11.** Isobaric  $\rho - T$  data obtained in the calculations for the empty sI clathrate hydrate phase.



**Figure S12.** Potential energy obtained from the calculations (symbols) and corresponding linear correlations (lines): ◀) 1 bar, ▶) 100 bar, ▲) 500 bar, ▼) 1000 bar, ■) 2000 bar and ●) 5000 bar for the empty sI clathrate hydrate phase.

**Table SI1.** Isobaric thermal expansivity maxima,  $\alpha_p$ , for the curves in Figure 3b.

| $P$ (bar)                             | 1     | 100   | 500   | 1000  | 2000  | 5000  |
|---------------------------------------|-------|-------|-------|-------|-------|-------|
| $T^{max}$ (K)                         | 194.7 | 193.0 | 193.8 | 189.6 | 183.0 | 166.2 |
| $\alpha_p$ ( $10^{-5}\text{K}^{-1}$ ) | 5.306 | 5.235 | 5.025 | 4.710 | 4.148 | 2.630 |



**Figure SI3.** Snapshots of two different views of the final configuration of the (5000 bar, 220 K) simulation after 50 ns reveal clathrate integrity. *top*) volumetric representation of the atomic clouds (not to scale) and *bottom*) ball and stick model for water (O – red, H – white).