

## **Supporting information**

### **Effect of chiral purity on adjuvanticity of archaeol-based glycolipids**

Sophie Régnier<sup>a</sup>, Edmond Lam<sup>a</sup>, Vinicio Vasquez<sup>a</sup>, Camilo F. Martinez-Farina<sup>b</sup>, , Felicity C. Stark<sup>c</sup>, Gerard Agbayani<sup>c</sup>, Lise Deschatelets<sup>c</sup>, Renu Dudani<sup>c</sup>, Blair A. Harrison<sup>c</sup>, Bassel Akache<sup>c</sup>, Michael J. McCluskie<sup>c\*</sup>, and Usha D. Hemraz<sup>a\*</sup>

<sup>a</sup>Aquatic and Crop Resource Development, National Research Council of Canada, 6100  
Royalmount Avenue, Montreal, Quebec, H4P 2R2, Canada

<sup>b</sup>Aquatic and Crop Resource Development Research Center, National Research Council  
of Canada, Halifax, Nova Scotia, B3H 3Z1, Canada

<sup>c</sup>Human Health Therapeutics, National Research Council of Canada, 1200 Montreal Rd,  
Ottawa, Ontario, K1A 0R6, Canada

## **Table of Contents**

1. Optical rotation	S3
2. Synthesis and characterization	S4-S6
3. LC-HRMS of archaeol	S7-S9
4. C8 reverse phase chromatography of archaeol	S10-S12
5. Chiral chromatography of archaeol (polar organic)	S13-S16
6. LC-HRMS of phytol	S17-S18
7. C8 reverse phase chromatography of phytol	S19
8. Chiral chromatography of archaeol and phytol (reverse phase)	S20-S25
9. Chiral chromatography (reverse phase) for elucidation of stereoisomers	S26
10. Percentage of archaeol 3 in 100% R configuration	S27-S28
11. NMR spectra of compounds	S29-S60
References	S61

## 1. Optical rotation

For comparative purposes, optical rotations were measured on a Rudolph Autopol I polarimeter for samples prepared in chloroform. The specific rotation  $[\alpha]_D^T$  values (Table S1) for archaeol and sucrose samples were obtained using a 100 mm polarimeter cell; D refers to the D-line of sodium (589 nm); Temperatures (T) are given in degrees Celsius (°C). Control and calibration experiments were also conducted using sucrose (Caledon, catalog # 8720-1, lot# 83129).

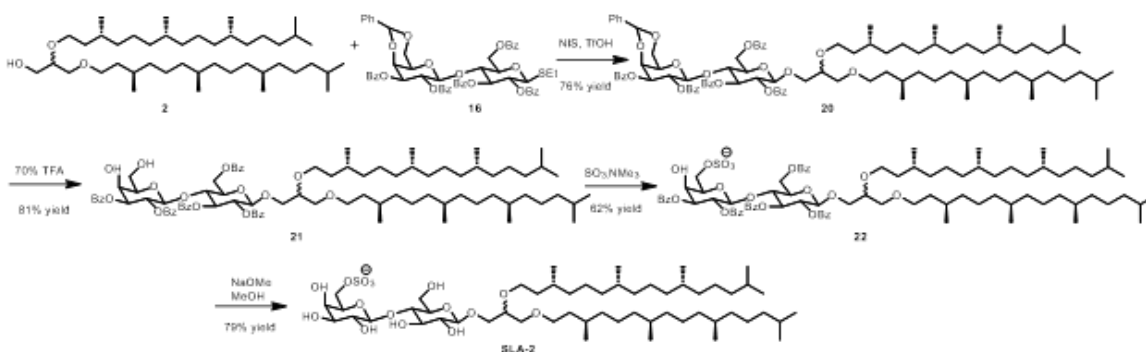
**Table S1. Specific rotation of archaeols and sucrose**

<b>Sample</b>	<b>Concentration (g/100 mL)</b>	<b>Specific rotation (°dm<sup>-1</sup>cm<sup>3</sup>g<sup>-1</sup>)</b>	<b>Temp. (°C)</b>
Archaeol <b>1</b>	4.96	9.68	24.6
Archaeol <b>2</b>	1.92	9.38	25.0
Archaeol <b>3</b>	0.84	0	25.0
Sucrose	20.02	66.28	25.6
Sucrose	10.01	68.33	25.5
Sucrose	5.01	67.66	25.2
Sucrose	1.00	69.00	25.2
Sucrose	0.50	70.00	25.1

## 2. Synthesis and characterization

SLA-1 and SLA-2 was synthesized according to a previously reported procedure that employs the bacterial archaeol 1.<sup>1-2</sup>

**Compound SLA-1.** <sup>1</sup>HNMR (1:1 CD<sub>3</sub>OD/CD<sub>2</sub>Cl<sub>2</sub>, 700 MHz)  $\delta$  4.32 (dd, 2H,  $J_1 = 2.8$  Hz,  $J_2 = 7.7$  Hz), 4.28 (brt, 1H,  $J_1 = 9.5$  Hz,  $J_2 = 10.9$  Hz), 4.09 (dd, 1H,  $J_1 = 3.5$  Hz,  $J_2 = 10.5$  Hz), 3.93 (dd, 1H,  $J_1 = 3.5$  Hz,  $J_2 = 10.5$  Hz), 3.87-3.81 (m, 4H), 3.56-3.60 (m, 8H), 3.41 (tt, 1H,  $J_1 = 3.5$  Hz,  $J_2 = 9.1$  Hz), 3.30 (m, 1H), 1.39-1.05 (m, 44H), 0.89-0.84 (m, 30H); <sup>13</sup>CNMR (1:1 CD<sub>3</sub>OD/CD<sub>2</sub>Cl<sub>2</sub>, 176 MHz)  $\delta$  105.0, 103.6, 83.0, 78.5, 75.5, 75.1, 74.3, 74.1, 73.7, 71.7, 70.7, 70.6, 69.7, 69.2, 69.2, 67.6, 62.0, 40.0, 38.1, 38.1, 38.0, 38.0, 38.0, 38.0, 37.9, 37.6, 37.3, 33.5, 33.5, 33.4, 30.5, 30.5, 28.6, 25.4, 25.1, 25.0, 22.9, 22.8, 20.1, 20.0, 20.0, 20.0 ppm; HRMS (ESI-) calcd. for C<sub>55</sub>H<sub>108</sub>O<sub>16</sub>S [M-Na+H]<sup>-</sup>: 1055.7285, found 1055.7276.  $[\alpha]_D^{23} = -1.8^\circ$  (c = 1.11, 1:1 CH<sub>2</sub>Cl<sub>2</sub>/MeOH). Anal. Calcd. For C<sub>55</sub>H<sub>107</sub>NaO<sub>16</sub>S (1078.72): C, 61.20; H, 9.99; N, 0.00, found C, 60.51; H, 9.83; N, 0.00.



**Scheme S1:** Synthesis of SLA-2

**Compound 20.** <sup>1</sup>HNMR (CDCl<sub>3</sub>, 700 MHz)  $\delta$  7.99 (d, 2H,  $J = 9.8$  Hz), 7.94-7.87 (m, 8H), 7.57 (t, 1H,  $J = 10.5$  Hz), 7.50-7.41 (m, 5H), 7.37-7.28 (m, 12H), 7.16 (t, 2H,  $J = 11.2$  Hz), 5.84 (t, 1H,  $J = 12.6$  Hz), 5.78 (dd, 1H,  $J_1 = 11.2$  Hz,  $J_2 = 14.7$  Hz), 5.34 (dd, 1H,  $J_1 = 10.5$  Hz,  $J_2 = 13.3$  Hz), 5.28 (s, 1H), 5.15 (dd, 1H,  $J_1 = 4.9$  Hz,  $J_2 = 14.7$  Hz), 4.83 (d, 1H,  $J = 11.2$  Hz), 4.75 (d, 1H,  $J = 11.2$  Hz), 4.61 (dd, 1H,  $J_1 = 2.8$  Hz,  $J_2 = 16.8$  Hz), 4.37 (dd, 1H,  $J_1 = 6.3$  Hz,  $J_2 = 16.8$  Hz), 4.30 (d, 1H,  $J = 4.9$  Hz), 4.22 (t, 1H,  $J = 13.3$  Hz), 3.86-3.83 (m, 2H), 3.77 (d, 1H,  $J = 16.1$  Hz), 3.57 (d, 1H,  $J = 14.7$  Hz), 3.51-3.44 (m, 2H), 3.35-

3.21 (m, 6H), 2.95 (s, 1H), 1.40-1.56 (m, 4H), 1.25-1.00 (m, 44H), 0.88-0.81 (m, 24H), 0.77 (d, 3H,  $J = 6.5$  Hz), 0.68 (d, 3H,  $J = 6.5$  Hz);  $^{13}\text{C}$ NMR ( $\text{CDCl}_3$ , 176 MHz)  $\delta$  166.3, 165.8, 165.4, 165.2, 165.0, 137.6, 133.5, 133.3, 133.2, 133.1, 130.0, 130.0, 129.8, 129.8, 129.7, 129.5, 129.0, 128.9, 128.5, 128.5, 128.4, 128.4, 128.4, 128.1, 126.5, 101.6, 101.1, 100.7, 77.8, 77.3, 77.2, 77.0, 76.9, 74.2, 73.2, 72.8, 72.7, 72.4, 70.7, 70.6, 70.1, 69.5, 69.2, 68.1, 66.5, 62.5, 39.5, 37.7, 37.6, 37.6, 37.6, 37.5, 37.4, 37.4, 37.0, 36.6, 33.0, 32.9, 30.0, 29.8, 28.1, 24.9, 24.6, 24.6, 24.4, 22.9, 22.8, 19.9, 19.8, 19.6 ppm; HRMS (ESI+) calcd. for  $\text{C}_{97}\text{H}_{132}\text{O}_{18}\text{Na}$   $[\text{M}+\text{Na}]^+$ : 1607.9306, found 1607.9445.

*Compound 21.*  $^1\text{H}$ NMR ( $\text{CDCl}_3$ , 700 MHz)  $\delta$  8.04 (d, 2H,  $J = 7.7$  Hz),  $\delta$  7.97 (d, 2H,  $J = 7.7$  Hz),  $\delta$  7.93 (d, 4H,  $J = 7.7$  Hz),  $\delta$  7.91 (d, 2H,  $J = 7.7$  Hz), 7.60 (t, 1H,  $J = 7.7$  Hz), 7.56 (t, 1H,  $J = 10.5$  Hz), 7.50-7.41 (m, 6H), 7.37-7.32 (m, 5H), 7.24 (t, 2H,  $J = 7.7$  Hz), 5.44 (t, 1H,  $J = 7.7$  Hz), 5.06 (dd, 1H,  $J_1 = 3.5$  Hz,  $J_2 = 10.5$  Hz), 4.76 (t, 2H,  $J = 7.7$  Hz), 4.59 (dd, 1H,  $J_1 = 1.4$  Hz,  $J_2 = 11.9$  Hz), 4.42 (dd, 1H,  $J_1 = 4.2$  Hz,  $J_2 = 11.9$  Hz), 4.19-4.17 (m, 2H), 3.88 (dd, 1H,  $J_1 = 2.8$  Hz,  $J_2 = 10.5$  Hz), 3.83 (m, 1H), 3.51-3.47 (m, 2H), 3.37-3.32 (m, 4H), 3.28-3.24 (m, 5H), 1.55-0.95 (m, 48H), 0.89-0.81 (m, 24H), 0.77 (d, 3H,  $J = 5$  Hz), 0.68 (d, 3H,  $J = 4.5$  Hz);  $^{13}\text{C}$ NMR (1:1  $\text{CD}_3\text{OD}/\text{CD}_2\text{Cl}_2$ , 176 MHz)  $\delta$  66.0, 165.9, 165.5, 165.3, 165.2, 133.7, 133.5, 133.4, 133.4, 133.3, 130.0, 130.0, 129.9, 129.9, 129.8, 129.8, 129.7, 129.6, 129.3, 129.0, 128.8, 128.6, 128.5, 128.5, 101.4, 101.2, 74.3, 74.1, 73.7, 72.9, 71.9, 70.7, 70.6, 70.1, 69.7, 69.3, 68.4, 62.7, 62.7, 39.5, 37.6, 37.6, 37.5, 37.5, 37.4, 37.4, 37.0, 36.6, 32.9, 32.9, 32.9, 30.0, 29.7, 28.1, 24.9, 24.6, 24.6, 24.5, 24.4, 22.9, 22.8, 22.8, 21.2, 19.9, 19.8, 19.8, 19.6 ppm; HRMS (ESI+) calcd. for  $\text{C}_{90}\text{H}_{128}\text{O}_{18}\text{Na}$   $[\text{M}+\text{Na}]^+$ : 1519.8993, found 1519.8977.

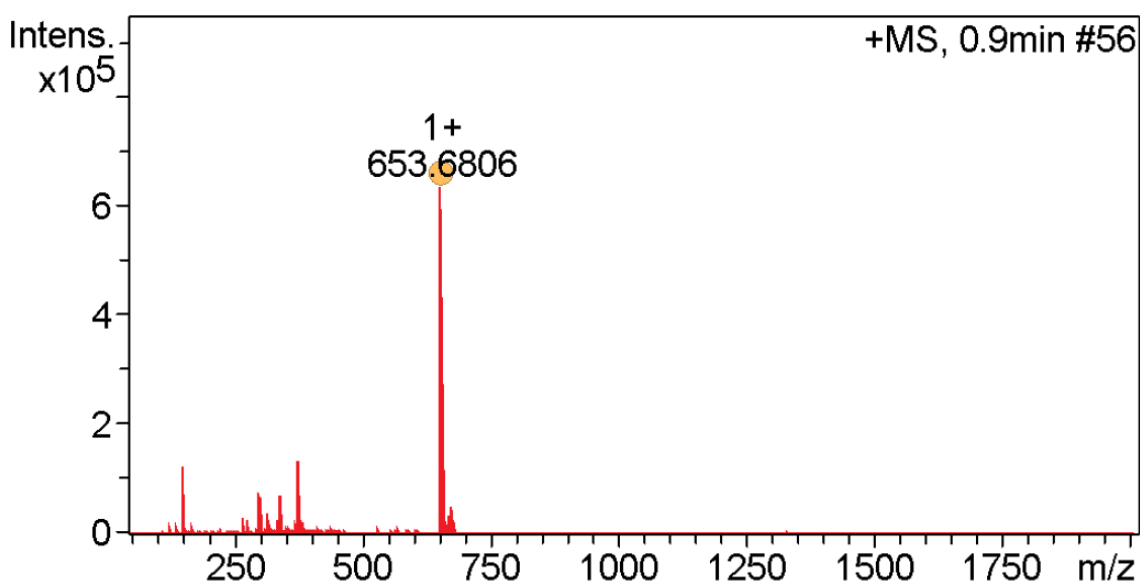
*Compound 22.*  $^1\text{H}$ NMR (1:1  $\text{CD}_3\text{OD}/\text{CD}_2\text{Cl}_2$ , 700 MHz)  $\delta$  8.00 (d, 2H,  $J = 7.7$  Hz), 7.94-7.91 (m, 6H), 8.00 (d, 2H,  $J = 7.7$  Hz), 7.86 (d, 2H,  $J = 7.7$  Hz), 7.55 (t, 1H,  $J = 7.7$  Hz), 7.52 (t, 1H,  $J = 7.7$  Hz), 7.49-7.43 (m, 5H), 7.38 (t, 2H,  $J = 7.7$  Hz), 7.32 (t, 3H,  $J = 7.7$  Hz), 7.23 (t, 2H,  $J = 7.7$  Hz), 5.71 (t, 1H,  $J = 9.1$  Hz), 5.62 (dd, 1H,  $J_1 = 7.7$  Hz,  $J_2 = 9.8$  Hz), 5.32 (m, 1H), 5.12 (dd, 1H,  $J_1 = 3.5$  Hz,  $J_2 = 10.5$  Hz), 4.82 (d, 1H,  $J = 7.7$  Hz), 4.74 (d, 1H,  $J = 7.7$  Hz), 4.58 (dd, 1H,  $J_1 = 1.4$  Hz,  $J_2 = 11.9$  Hz), 4.41 (dd, 1H,  $J_1 = 4.9$  Hz,  $J_2 = 11.9$  Hz), 4.21 (t, 1H,  $J = 9.1$  Hz), 4.15 (d, 1H,  $J = 2.8$  Hz), 3.83 (m, 2H), 3.57 (d, 3H),

3.45 (m, 1H), 3.42 (m, 1H), 3.31 (m, 2H), 3.20 (m, 4H), 1.50 (m, 4H), 1.42-0.81 (m, 68H); 0.75 (d, 3H,  $J = 7.0$  Hz), 0.68 (d, 3H,  $J = 6.3$  Hz);  $^{13}\text{C}$ NMR (1:1  $\text{CD}_3\text{OD}/\text{CD}_2\text{Cl}_2$ , 176 MHz)  $\delta$  166.4, 166.4, 166.2, 165.9, 165.7, 134.0, 133.7, 133.7, 133.6, 130.0, 130.0, 129.9, 129.9, 129.9, 129.7, 129.6, 129.2, 128.9, 128.8, 128.8, 128.7, 101.8, 101.4, 77.9, 76.9, 74.4, 73.8, 73.3, 73.0, 72.5, 70.7, 70.6, 70.4, 70.2, 69.2, 65.8, 64.1, 62.9, 39.7, 37.8, 37.8, 37.8, 37.7, 37.7, 37.7, 37.7, 37.6, 37.6, 37.2, 36.8, 33.1, 30.2, 30.0, 28.3, 25.1, 24.8, 24.8, 24.7, 24.6, 19.8, 19.8, 19.7, 19.6 ppm HRMS (ESI-) calcd, for  $\text{C}_{90}\text{H}_{128}\text{O}_{21}\text{S}$   $[\text{M}-\text{Na}+\text{H}]^-$ : 1575.8596, found 1575.9055.

*Compound SLA-2.*  $^1\text{H}$ NMR (1:1  $\text{CD}_3\text{OD}/\text{CD}_2\text{Cl}_2$ , 700 MHz)  $\delta$  4.33 (brt, 2H,  $J1 = 6.3$  Hz,  $J2 = 7.7$  Hz), 4.27 (brt, 1H,  $J1 = 9.1$  Hz,  $J2 = 10.5$  Hz), 4.11 (dd, 1H,  $J1 = 4.2$  Hz,  $J2 = 11.2$  Hz), 3.93 (dd, 1H,  $J1 = 4.2$  Hz,  $J2 = 10.5$  Hz), 3.88-3.84 (m, 4H), 3.67-3.48 (m, 12H), 3.42-3.41 (m, 1H), 3.31-3.30 (m, 1H), 1.64-1.05 (m, 44H), 0.90-0.84 (m, 30H);  $^{13}\text{C}$ NMR (1:1  $\text{CD}_3\text{OD}/\text{CD}_2\text{Cl}_2$ , 175 MHz)  $\delta$  105.1, 103.8, 82.8, 78.8, 78.7, 75.7, 75.4, 74.4, 74.1, 73.9, 71.8, 71.1, 70.7, 70.7, 69.8, 69.8, 69.5, 69.4, 69.3, 67.7, 62.0, 40.2, 38.2, 38.2, 38.2, 38.2, 38.1, 38.1, 38.1, 38.0, 37.7, 37.4, 33.6, 33.6, 33.6, 30.6, 30.6, 28.8, 25.6, 25.2, 25.2, 25.2, 25.2, 25.2, 23.0, 22.9, 20.1, 20.1, 20.1, 20.1, 20.0, 19.9 ppm; HRMS (ESI-) calcd. for  $\text{C}_{55}\text{H}_{108}\text{O}_{16}\text{S}$   $[\text{M}-\text{Na}+\text{H}]^-$ : 1055.7285, found 1055.7282.  $[\alpha]_{\text{D}}^{23} = -2.67^\circ$  ( $c = 1.87$ , 1:1  $\text{CHCl}_3/\text{MeOH}$ ). Anal. Calcd. For  $\text{C}_{55}\text{H}_{107}\text{NaO}_{16}\text{S}$  (1078.72): C, 61.20; H, 9.99; N, 0.00, found C, 60.67; H, 9.82; N, 0.00.

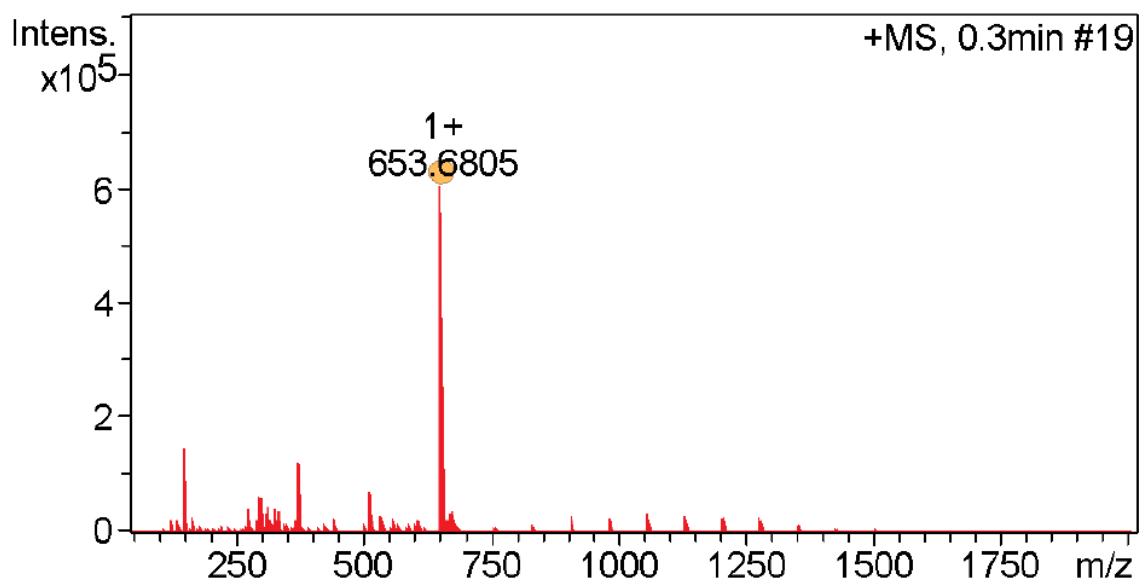
### 3. LC-HRMS analysis of archaeol

Prior to conducting the chiral HPLC analysis, HRMS was used to confirm the identity of the compounds. The latter were analyzed by infusion, in which the sample (e.g., archaeol at 10 µg/L in methanol) is infused into the mobile phase flow and passes directly into the mass spectrometer (TOF). Ammonium acetate (5 mM) and methanol were used as mobile phases at a flow of 0.1 mL/min. The compounds were identified using the exact mass of the molecular ion  $[M+H]^+$  and the isotopic distribution. The mass spectrometer was set to full MS scan from 100 m/z to 1500 m/z and operated in positive ionization mode. Archaeol has a chemical formula of  $C_{43}H_{88}O_3$  and exact mass of 652.6733; The  $[M+H]^+$  ion has an exact mass of 653.6806.



#	m/z	Res.	S/N	I	I %	FWHM
1	653.6806	9105	5171.4	634687	100.0	0.0718
2	654.6837	9348	2540.8	311469	49.1	0.0700
3	655.6859	9795	653.9	80137	12.6	0.0669
4	656.6888	10054	107.3	13148	2.1	0.0653

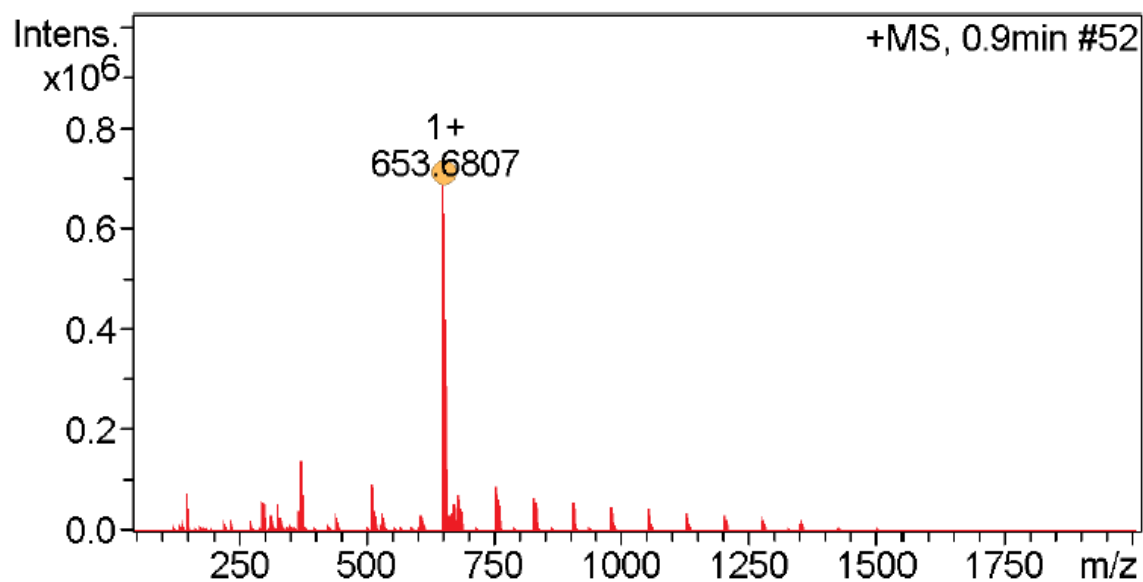
**Figure S1.** Mass spectrum of archaeol **1** (100% R) showing the  $[M+H]^+$  ion at m/z 653.6806 and the expected isotopic ratio (theoretical pattern: 653.6806 (100.0%); 654.6840 (47.6%); 655.6873 (11.7%); 656.6905 (2.0%)).



#	m/z	Res.	S/N	I	I %	FWHM
1	653.6805	9275	3808.2	605108	100.0	0.0705
2	654.6837	9512	1829.6	290699	48.0	0.0688
3	655.6855	9878	426.5	67763	11.2	0.0664
4	656.2196	7106	7.9	1252	0.2	0.0923
5	656.6892	9821	74.6	11858	2.0	0.0669
6	657.2215	8760	5.4	851	0.1	0.0750
7	657.6766	5605	10.2	1622	0.3	0.1173

**Figure S2.** Mass spectrum of archaeol 2 (94% R) showing the  $[M+H]^+$  ion at  $m/z$  653.6805 and the expected isotopic ratio (theoretical pattern: 653.6806 (100.0%); 654.6840 (47.6%); 655.6873 (11.7%); 656.6905 (2.0%)).



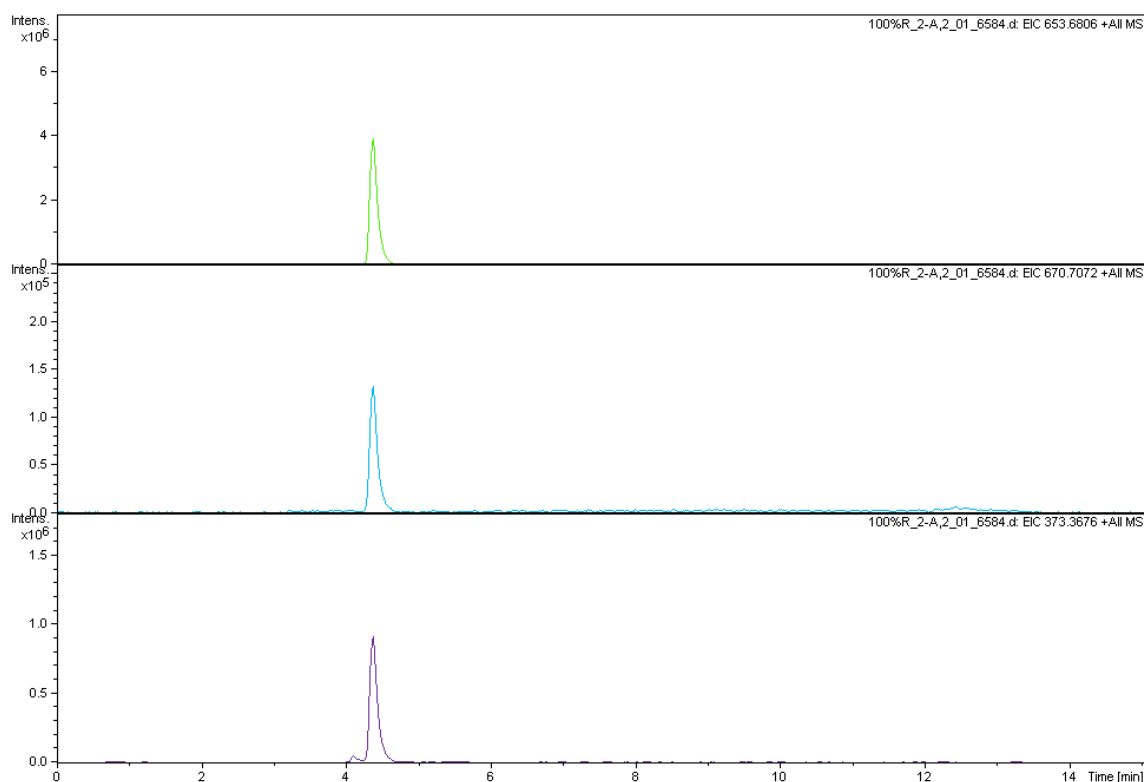


#	m/z	Res.	S/N	I	I %	FWHM
1	653.6807	9216	4190.8	684002	100.0	0.0709
2	654.2355	3377	16.3	2660	0.4	0.1937
3	654.6834	9459	2020.8	329820	48.2	0.0692
4	655.1680	3363	8.4	1377	0.2	0.1948
5	655.6859	9698	488.3	79701	11.7	0.0676
6	656.2209	8817	15.2	2487	0.4	0.0744
7	656.6875	9521	82.4	13441	2.0	0.0690
8	657.2297	10261	9.8	1599	0.2	0.0640
9	657.6929	9136	12.3	2000	0.3	0.0720

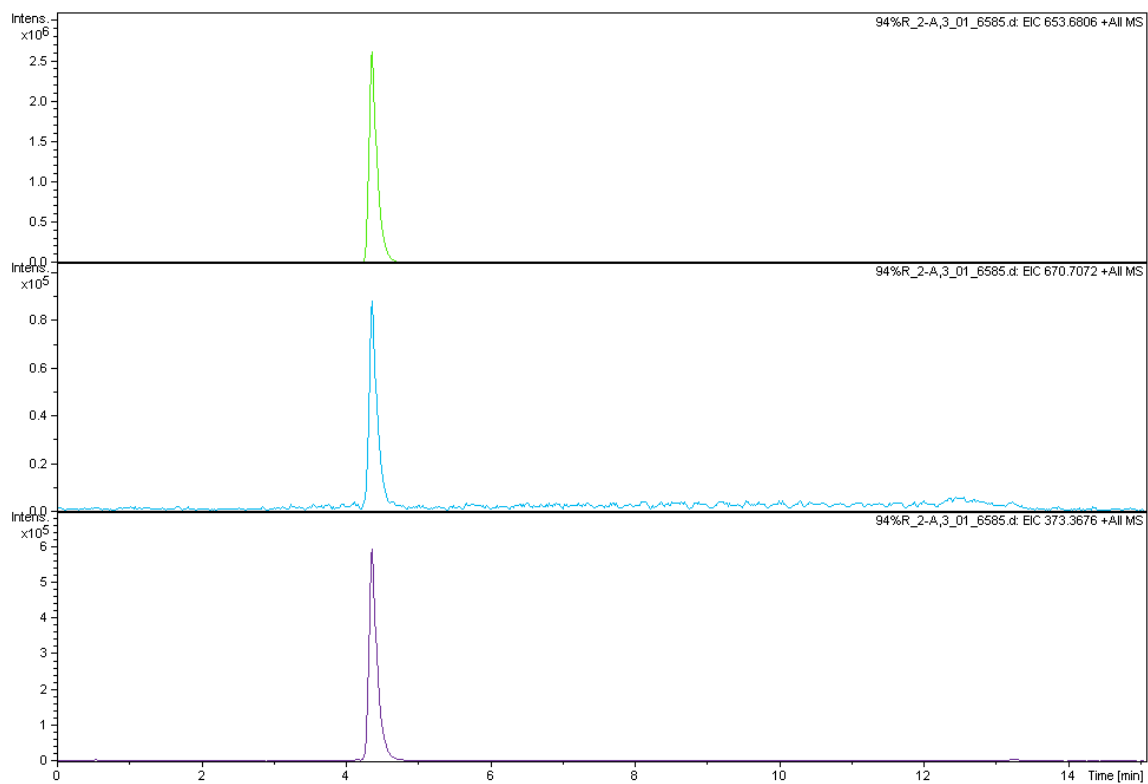
**Figure S3.** Mass spectrum of archaeol **3** (mixture of stereoisomers) showing the  $[M+H]^+$  ion at  $m/z$  653.6807 and the expected isotopic ratio (theoretical pattern: 653.6806 (100.0%); 654.6840 (47.6%); 655.6873 (11.7%); 656.6905 (2.0%)).

#### 4. Archaeol C8 reverse-phase chromatography for identification confirmation

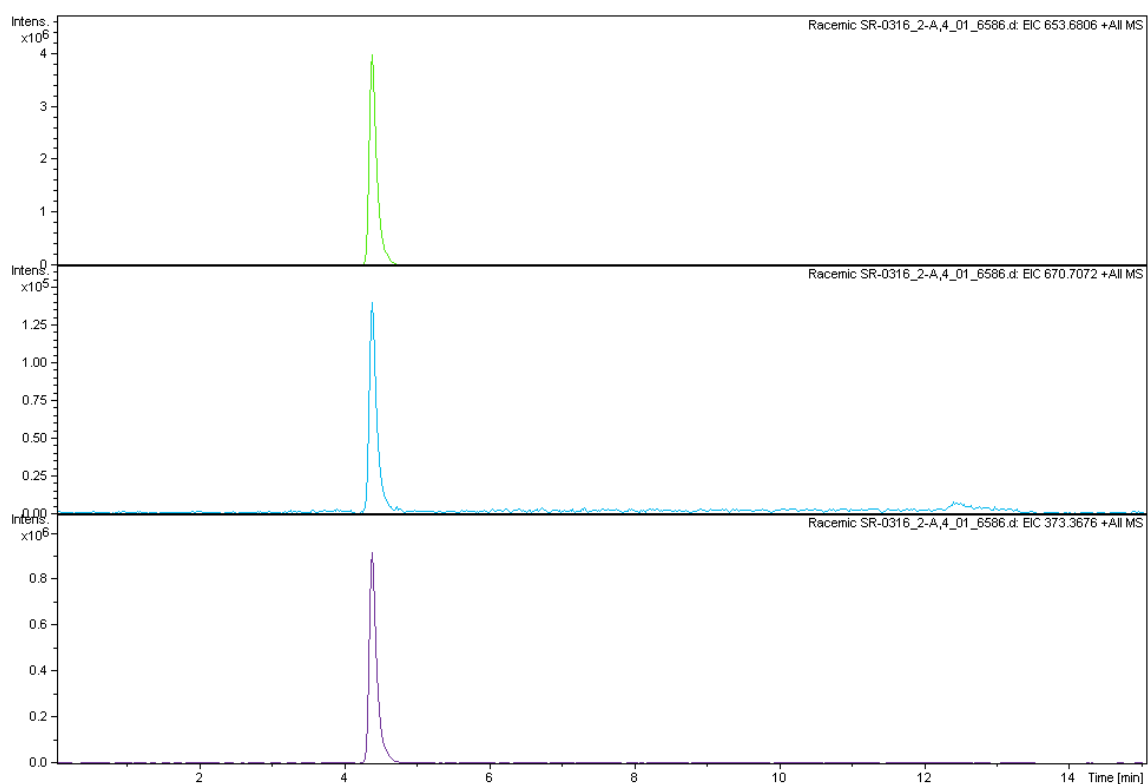
The compounds were also characterized by reverse-phase chromatography on a C8 column Halo C8 (50 x 3.0 mm, particle size 2.7  $\mu\text{m}$ ) from Advance Materials Technology) for identity confirmation. Ammonium acetate (5 mM) and methanol were used as mobile phases at a flow of 0.5 mL/min. A gradient of methanol from 95% to 100% within 2 min was used to elute the compound. The mass spectrometer (TOF) was set to full MS scan from 100 m/z to 1500 m/z and operated in positive ionization mode. The molecular ion  $[\text{M}+\text{H}]^+$  at m/z 653.6806, the ammonium adduct  $[\text{M}+\text{NH}_4]^+$  at m/z 670.7072, and a fragment at m/z 373.3676 corresponding to the loss of a phytol unit, were monitored to confirm the identity of compound. The same retention times (4.3 min) were obtained for the molecular ion 653.6806 m/z, ammonium adduct 670.7072 m/z and fragment 373.3676 m/z for the three archaeol samples.



**Figure S4.** Extracted ion chromatograms (m/z 653.6806, 670.7072 and 373.3676) for archaeol 1 (100% R).



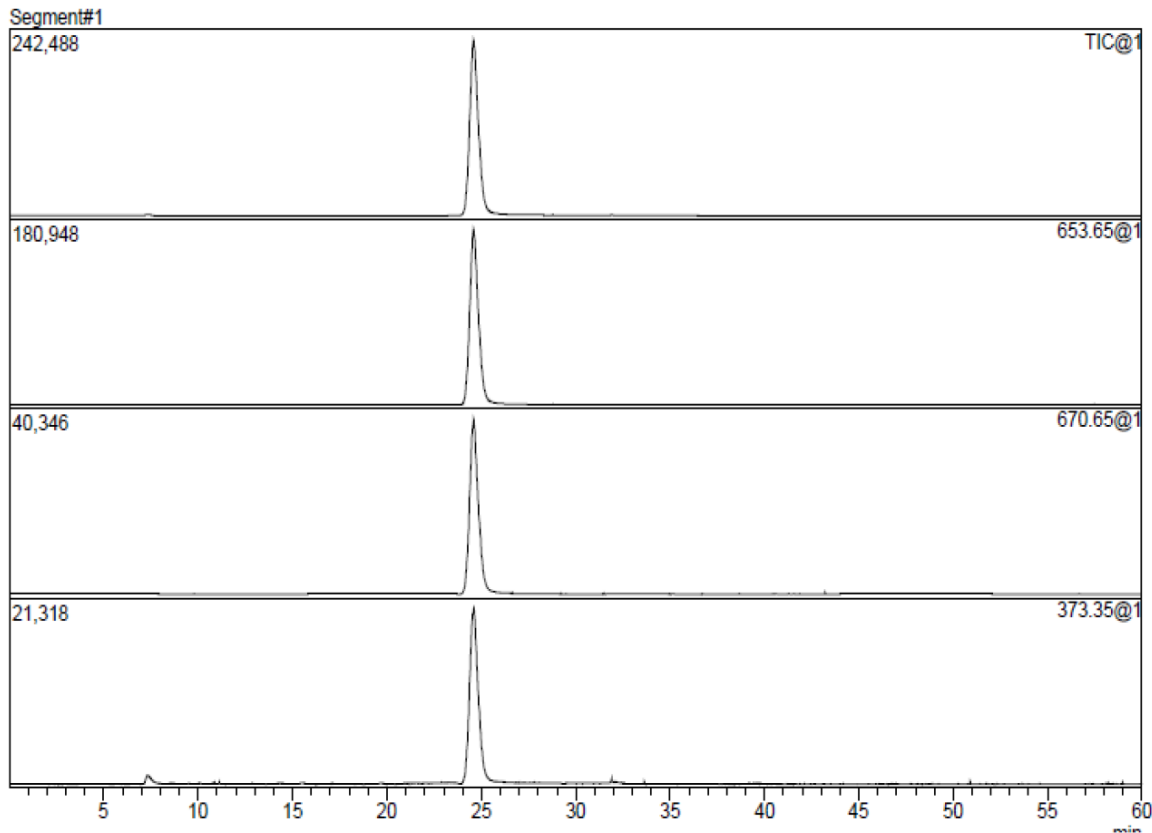
**Figure S5.** Extracted ion chromatograms (m/z 653.6806, 670.7072 and 373.3676) for archaeol 2 (94% R).



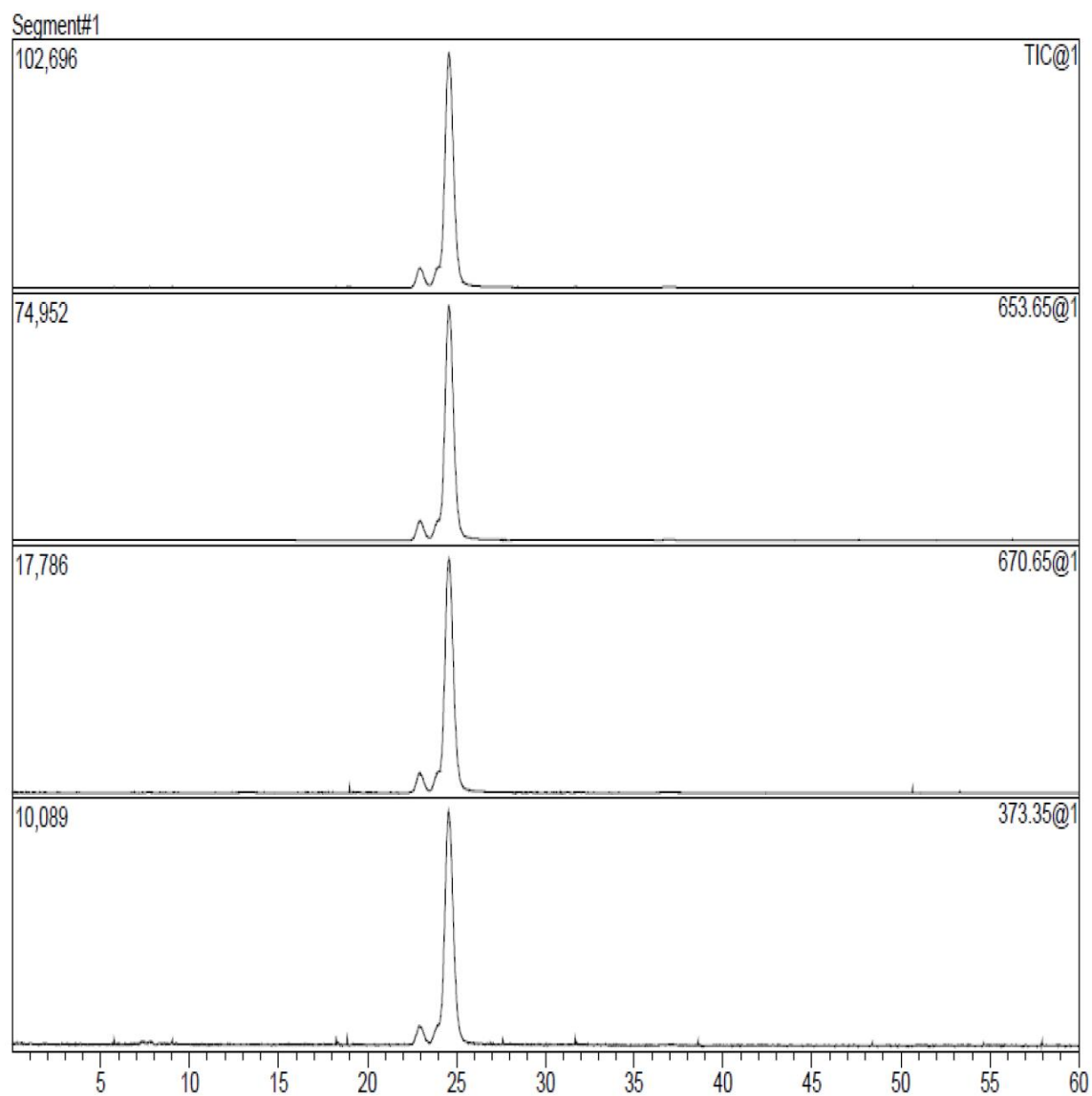
**Figure S6.** Extracted ion chromatograms ( $m/z$  653.6806, 670.7072 and 373.37) for archaeol **3** (mixture of stereoisomers).

## 5. Chiral chromatography (polar organic) for elucidation of stereoisomers

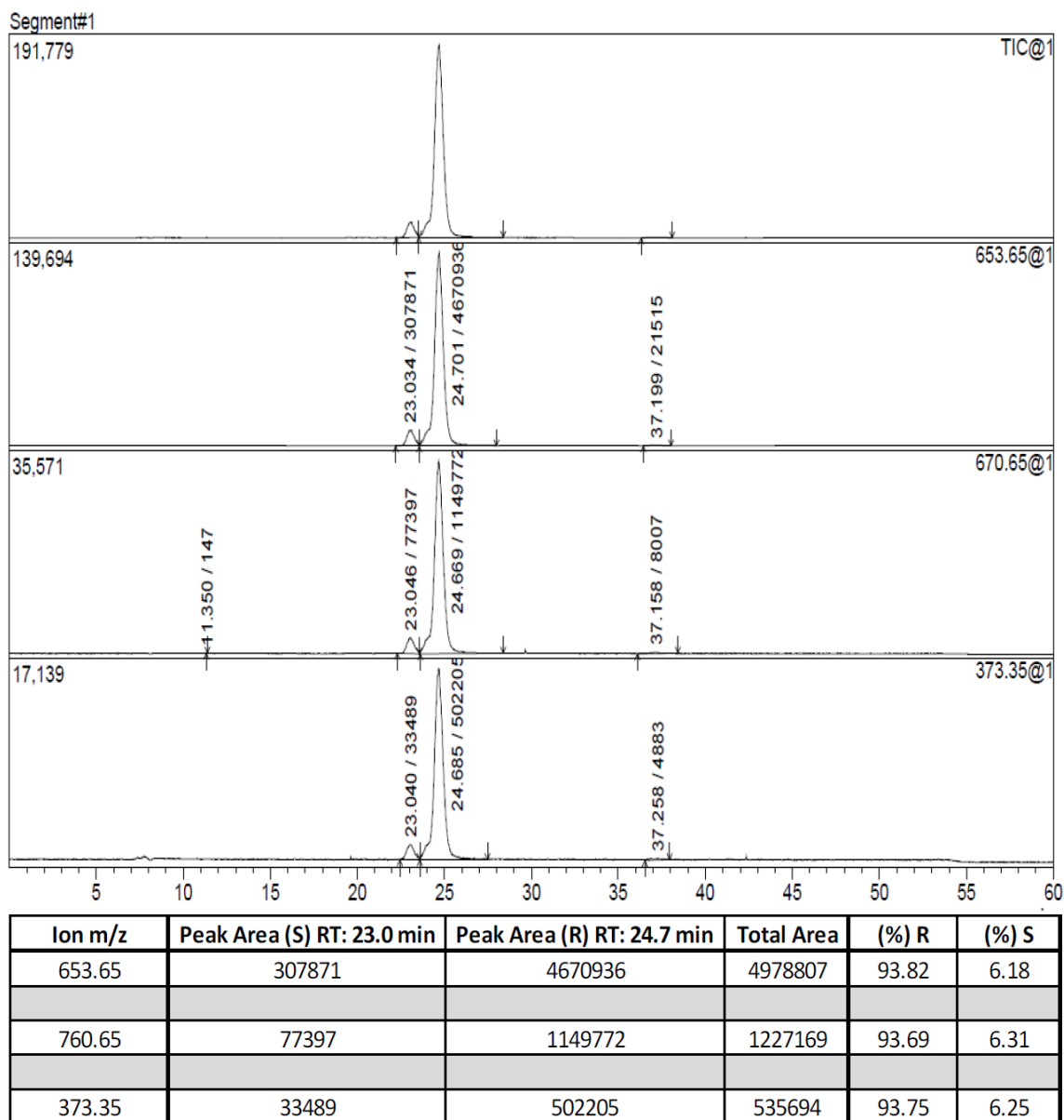
The elucidation of the various stereoisomers in the archaeol samples was attempted on a Lux immobilized phase chiral column (amylose tris (3- chloro-5-methylphenylcarbamate), Lux i-Amylose-3 (250 x 4.6 mm, particle size 3  $\mu$ m) from Phenomenex Inc.) Tests were performed in isocratic mobile phase conditions (not all conditions shown). A single quadrupole mass spectrometer was used for detection and was set to Selected Ion Mass (SIM) and operated in positive ionization mode. The molecular ion  $[M+H]^+$  at  $m/z$  653.65, the ammonium adduct  $[M+NH_4]^+$  at  $m/z$  670.65, and a fragment at  $m/z$  373.35 corresponding to the loss of a phytyl unit, were monitored to confirm the identity of compound.



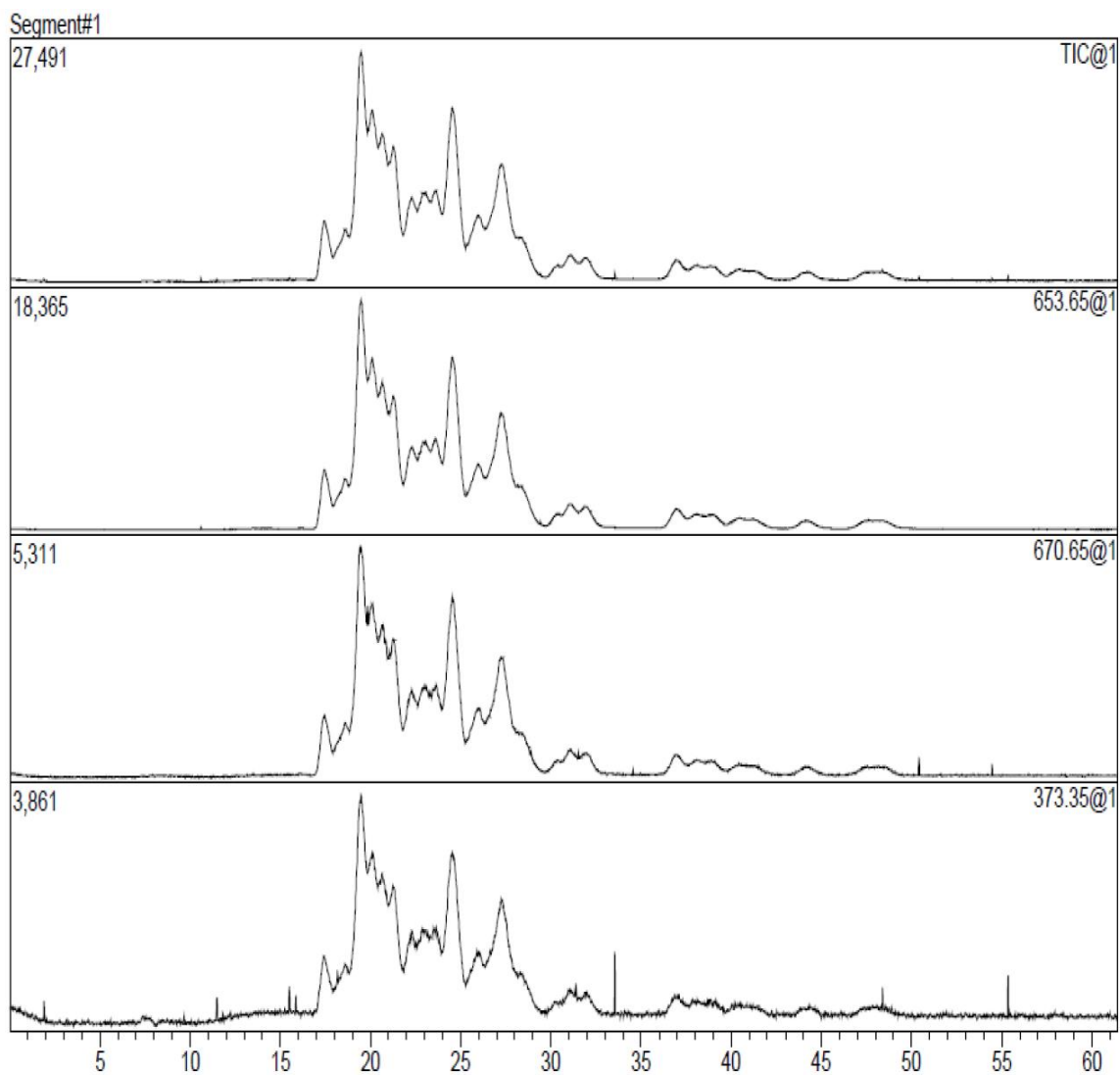
**Figure S7.** Total ion and extracted ion ( $m/z$  653.65, 670.65 and 373.35) chromatograms for archaeol **1** (100% R) using chiral chromatography in polar organic mode. Methanol was used as mobile phases at a flow of 0.4 mL/min with a post-column addition of 5 mM ammonium acetate at a flow of 0.1 mL/min.



**Figure S8.** Total ion and extracted ion ( $m/z$  653.65, 670.65 and 373.35) chromatograms for archaeol **2** (94% R) using chiral chromatography in polar organic mode. Methanol was used as mobile phases at a flow of 0.4 mL/min with a post-column addition of 5 mM ammonium acetate at a flow of 0.1 mL/min.



**Figure S9.** HPLC chiral purity of R and S isomers of archaeol **2** using chiral chromatography in polar organic mode. Methanol was used as mobile phases at a flow of 0.4 mL/min with a post-column addition of 5 mM ammonium acetate at a flow of 0.1 mL/min.

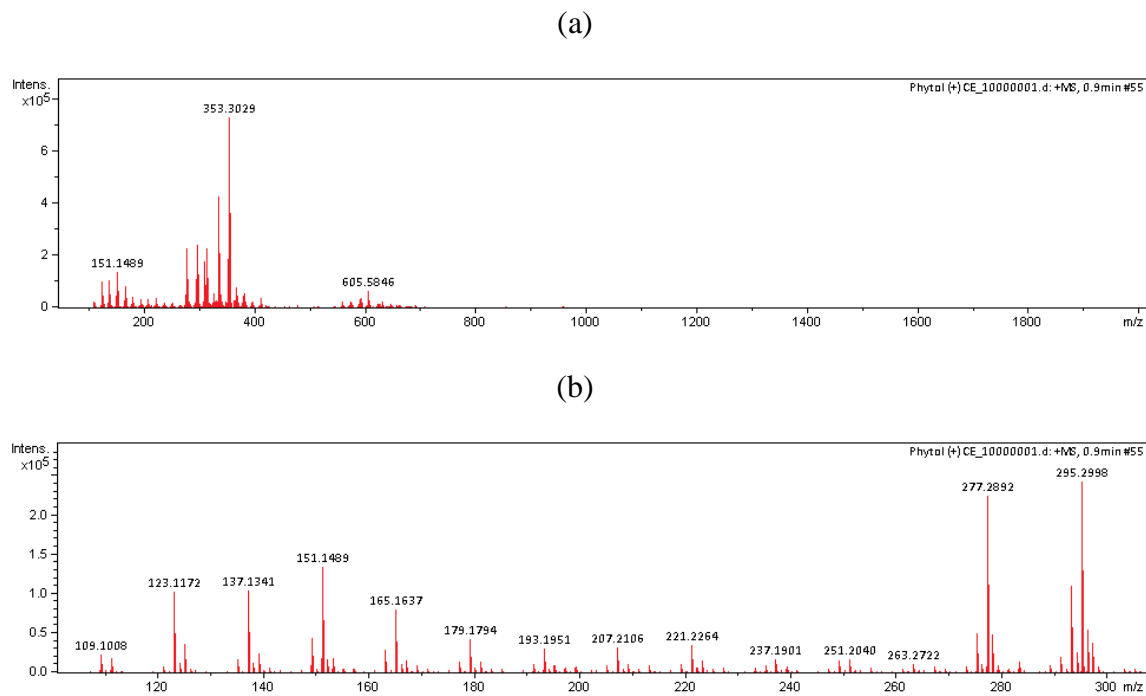


**Figure S10.** Total ion and extracted ion ( $m/z$  653.65, 670.65 and 373.35) chromatograms for archaeol **3** (mixture of stereoisomers) using chiral chromatography in polar organic mode. Methanol was used as mobile phases at a flow of 0.4 mL/min with a post-column addition of 5 mM ammonium acetate at a flow of 0.1 mL/min.



## 6. LC-HRMS analysis of phytol

Given that the extracted ion chromatogram of archaeol **3** showed a highly complex chiral composition with more than 20 peaks showing the same molecular ion  $[M+H]^+$  at  $m/z$  653.65, the analysis of phytol **7** by HRMS was done. The latter was analyzed by infusion, in which the sample is infused into the mobile phase flow and passes directly into the mass spectrometer (TOF). Ammonium acetate (5 mM) and methanol were used as mobile phases at a flow of 0.1 mL/min. The mass spectrometer was set to full MS scan from 100  $m/z$  to 1500  $m/z$  and operated in positive ionization mode. Phytol has a chemical formula of  $C_{20}H_{40}O$  and exact mass of 296.3079; The  $[M+H]^+$  ion has an exact mass of 297.3152.



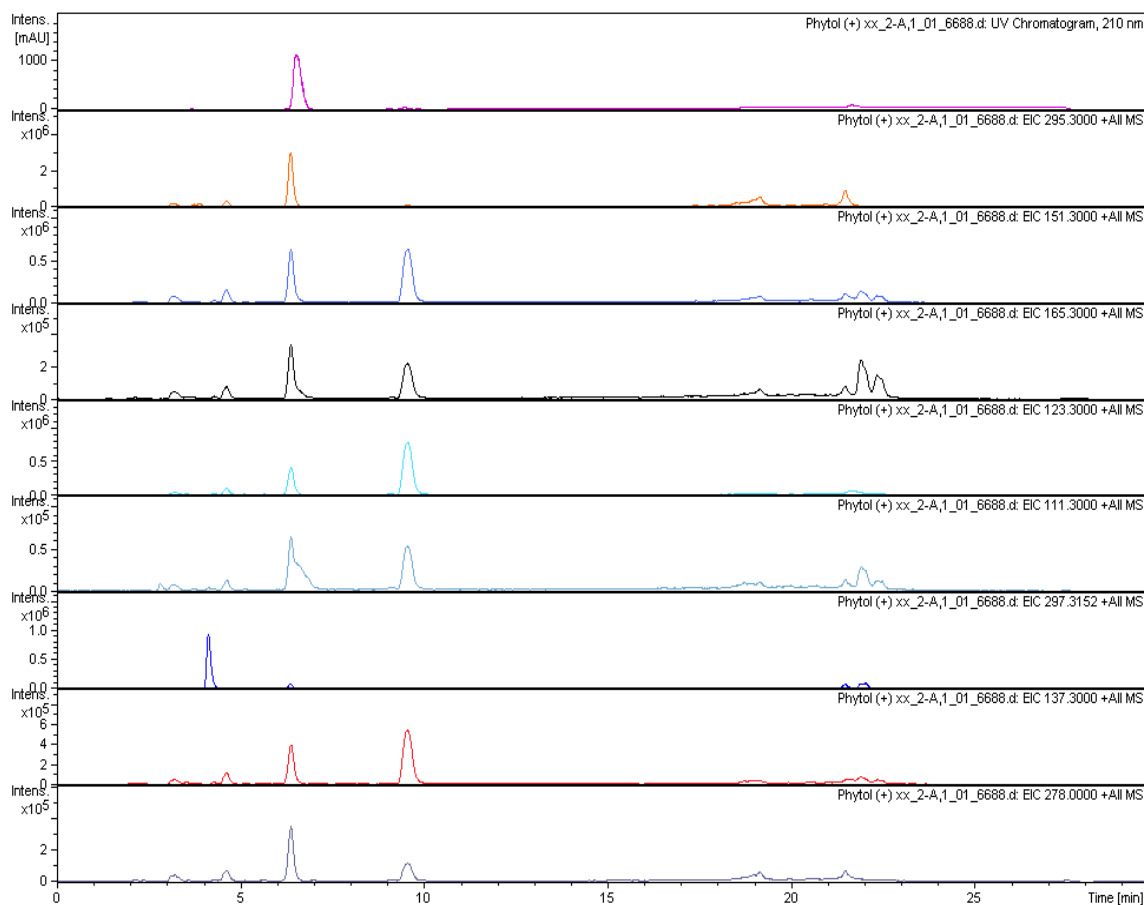
**Figure S11.** Mass spectrum of phytol **7** (97% mixture of isomers) showing the full spectrum (a) and a zoomed area between 100-300  $m/z$  (b). The molecular ion  $[M+H]^+$  ion at  $m/z$  297.3152 has a low intensity and fragments are generated rapidly.

#	m/z	Res.	S/N	I	I %	FWHM
1	109.1008	4831	1116.6	21875	9.1	0.0226
2	111.1162	4699	898.3	17906	7.5	0.0236
3	123.1172	5165	4627.3	100774	41.9	0.0238
4	125.1324	5011	1634.1	35720	14.9	0.0250
5	135.1183	5237	714.1	17265	7.2	0.0258
6	137.1341	5277	4252.9	103294	43.0	0.0260
7	139.1470	5084	963.8	23485	9.8	0.0274
8	149.1334	5762	1702.2	44376	18.5	0.0259
9	151.1489	5502	5098.9	133324	55.5	0.0275
10	152.1522	5805	641.8	16859	7.0	0.0262
11	153.1609	5756	685.5	18026	7.5	0.0266
12	163.1479	6028	1040.5	29257	12.2	0.0271
13	165.1638	6091	2789.5	79544	33.1	0.0271
14	179.1792	5931	1241.8	41635	17.3	0.0302
15	193.1952	6306	801.4	30372	12.6	0.0306
16	207.2107	6562	731.3	31305	13.0	0.0316
17	221.2263	6649	733.8	34135	14.2	0.0333
18	223.1756	6192	307.4	14373	6.0	0.0360
19	237.1900	5894	332.6	16981	7.1	0.0402
20	249.2570	7105	278.4	14662	6.1	0.0351
21	251.2039	6797	315.0	16657	6.9	0.0370
22	275.2735	7151	858.3	49155	20.5	0.0385
23	277.2892	7345	3856.5	222413	92.6	0.0378
24	278.2926	7270	834.7	48215	20.1	0.0383
25	291.2751	5249	325.2	19515	8.1	0.0555
26	293.2843	7650	1815.7	110348	45.9	0.0383
27	294.2938	5681	424.2	25817	10.7	0.0518
28	295.2999	7242	3948.3	240264	100.0	0.0408
29	296.3054	6685	880.4	53533	22.3	0.0443
30	297.3130	6980	597.4	36478	15.2	0.0426

**Figure S12.** Phytol **7** (97% mixture of isomers) showing the 30 most abundant peaks within the mass spectrum between 100-300 m/z. The molecular ion  $[M+H]^+$  ion at m/z 297.3152 has a low intensity and fragments are generated rapidly. Most of the fragments reported by NIST and NIH can be seen; m/z 111, 123, 137, 151, 165, 179, 223, 249, 278, 295, and 296.

## 7. Phytol C8 reverse-phase chromatography for identification confirmation

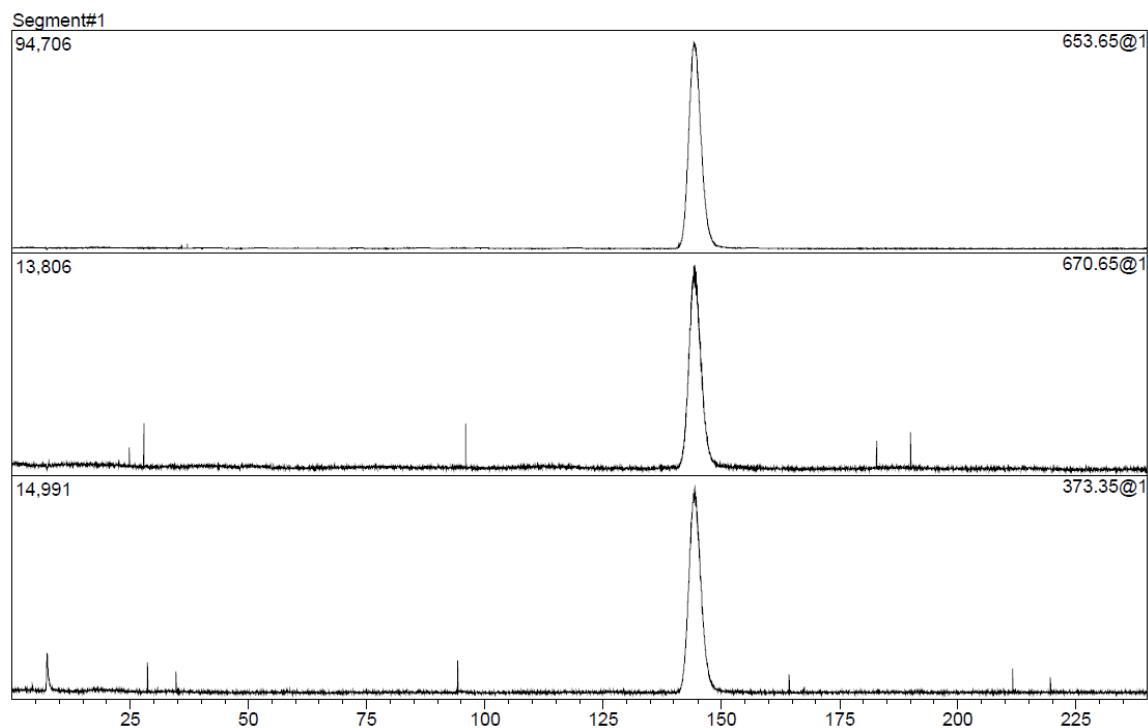
Phytol **7** (97% mixture of isomers) was also characterized by reverse-phase chromatography on a C8 column for identity confirmation. Ammonium acetate (5 mM) and methanol were used as mobile phases at a flow of 0.5 mL/min. A gradient of methanol from 80% to 100% within 10 min was used to achieve peak separation and elute the compound. The DA detector was set from 190-950 nm and 210 nm was selected to monitored and confirm the identity of the compound.<sup>3</sup> The mass spectrometer (TOF) was set to full MS scan from 100 m/z to 1500 m/z and operated in positive ionization mode. The molecular ion  $[M+H]^+$  at m/z 297.3152, reported fragments 295, 278, 165, 151, 123 and 111 were also monitored to confirm the identity of compound.<sup>4</sup> The same retention times (6.4 min) were obtained for the UV chromatogram, molecular ion and selected fragments.



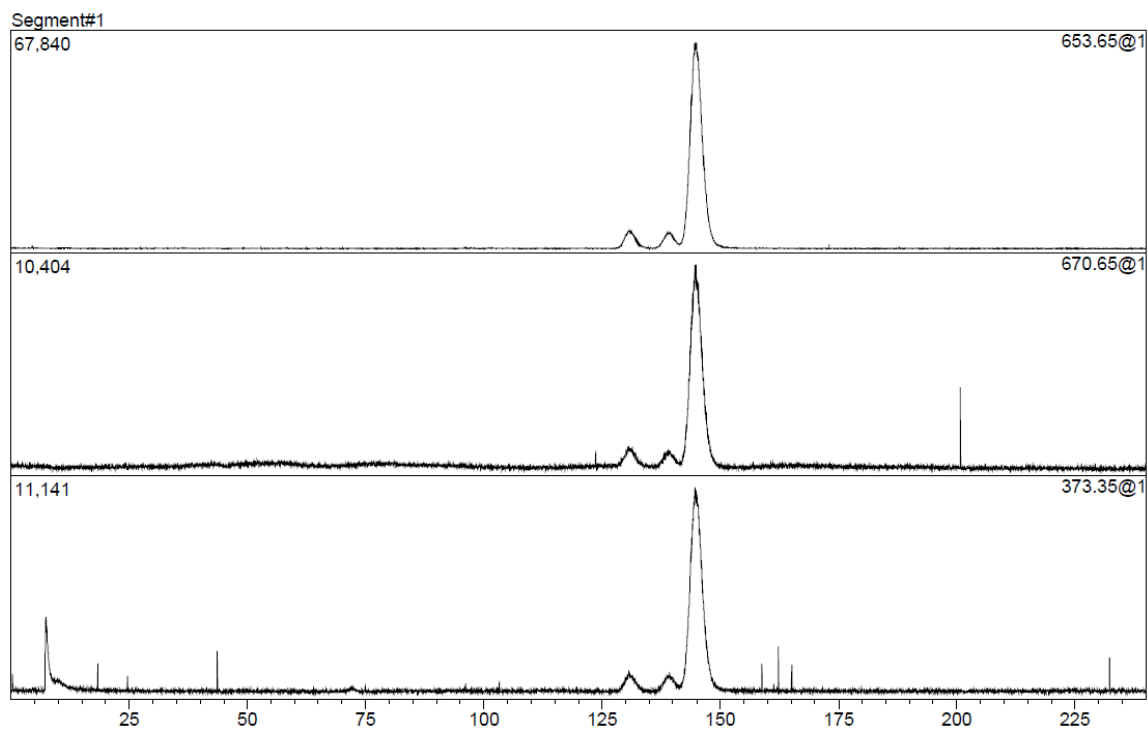
**Figure S13.** UV chromatogram and extracted ion chromatograms ( $m/z$  297.3152 and fragments 295, 278, 165, 151, 137, 123 and 111) for phytol **7**.

### **8. Chiral chromatography (reverse phase) for elucidation of stereoisomers**

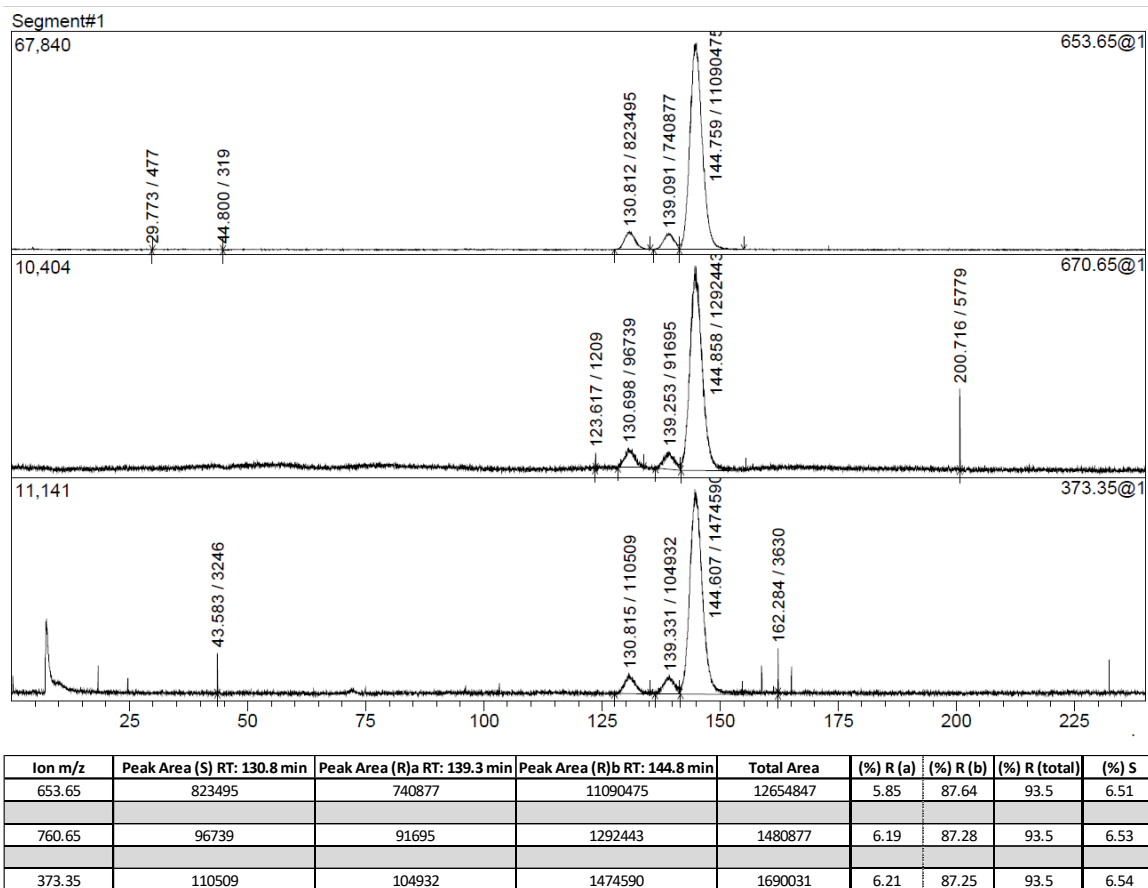
The elucidation of the various stereoisomers in the archaeol samples was attempted on a Lux immobilized phase chiral column (amylose tris (3-chloro-5-methylphenylcarbamate)). Phytol **7** (97% mixture of isomers) was also analyzed and the molecular ion  $[M+H]^+$  at  $m/z$  297.3152, reported fragments 295, 278, 165, 151, 123 and 111, were also monitored. Tests were performed in isocratic mobile phase conditions (not all conditions shown). A single quadrupole mass spectrometer was used for detection and was set to selected ion mass (SIM) and operated in positive ionization mode. The molecular ion  $[M+H]^+$  at  $m/z$  653.65, the ammonium adduct  $[M+NH_4]^+$  at  $m/z$  670.65, and a fragment at  $m/z$  373.35 corresponding to the loss of a phytyl unit, were monitored to confirm the identity of archaeol samples. Better resolution was obtained than the polar organic chromatography showing the shouldering minor peak of the R isomer (Figure S16). This confirms that within the 94% of (*R*)-2,3-bis(((3*R*,7*R*,11*R*)-3,7,11,15-tetramethylhexadecyl)oxy) propan-1-ol, there is a 6% of a different configuration of the chiral centers located on the phytanyl portion of the molecule, most probably at C3 of the phytyl unit. Again, as seen for the polar organic chromatography, a highly complex chiral composition can be noticed with more than 20 peaks showing the same molecular ion  $[M+H]^+$  at  $m/z$  653.65 but with better peak separation. The optically inactive archaeol **3**, with its 7 chiral centers, could have up to 128 enantiomer configurations ( $2^7 = 128$ ), which explains the numerous peaks observed (Figure S17). The chromatograms for phytol **7** show more than 15 peaks with the same molecular ion  $[M+H]^+$  at  $m/z$  297.3 and reported fragments (Figure S18). This correlates with the highly complex chiral composition observed in the chiral chromatography of optically inactive archaeol **3** (Figure S17).



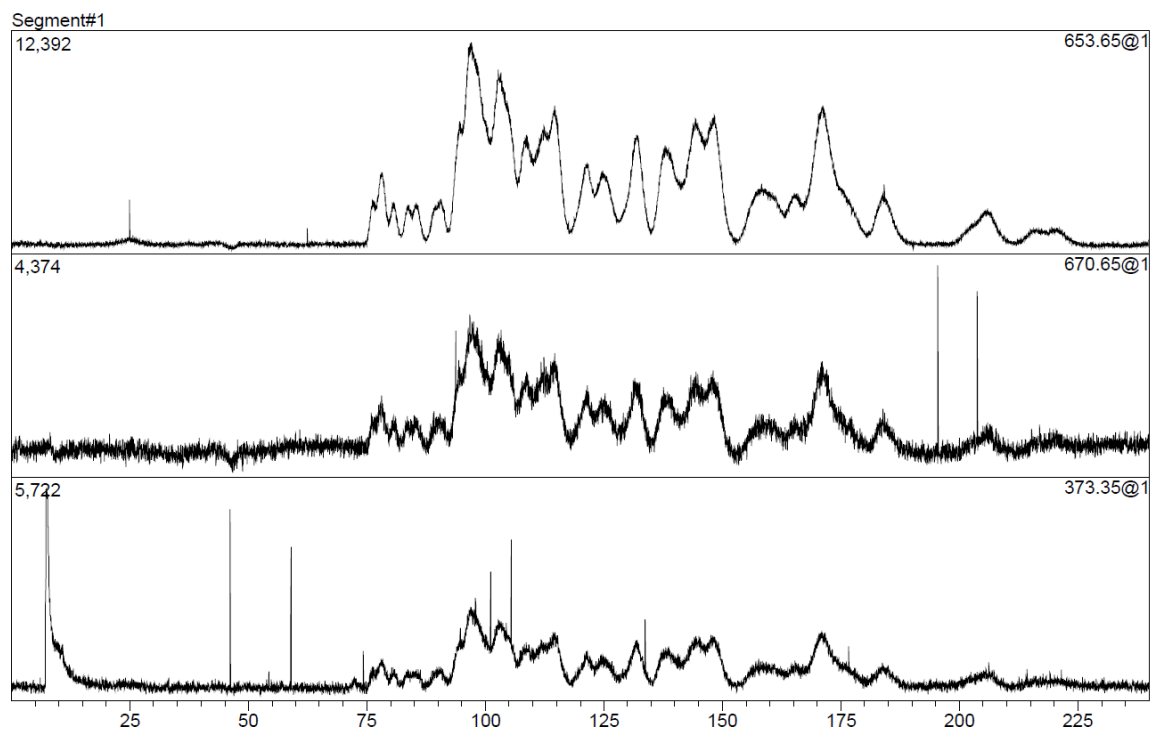
**Figure S14.** Total ion and extracted ion ( $m/z$  653.65, 670.65 and 373.35) chromatograms for archaeol **1** (100% R) using chiral chromatography in polar organic mode. Methanol:H<sub>2</sub>O 95:5% v/v was used as mobile phases at a flow of 0.4 mL/min with a post-column addition of 5 mM ammonium acetate at a flow of 0.1 mL/min.



**Figure S15.** Total ion and extracted ion ( $m/z$  653.65, 670.65 and 373.35) chromatograms for archaeol **2** (94% R) using chiral chromatography in polar organic mode. Methanol:H<sub>2</sub>O 95:5% v/v was used as mobile phases at a flow of 0.4 mL/min with a post-column addition of 5 mM ammonium acetate at a flow of 0.1 mL/min.

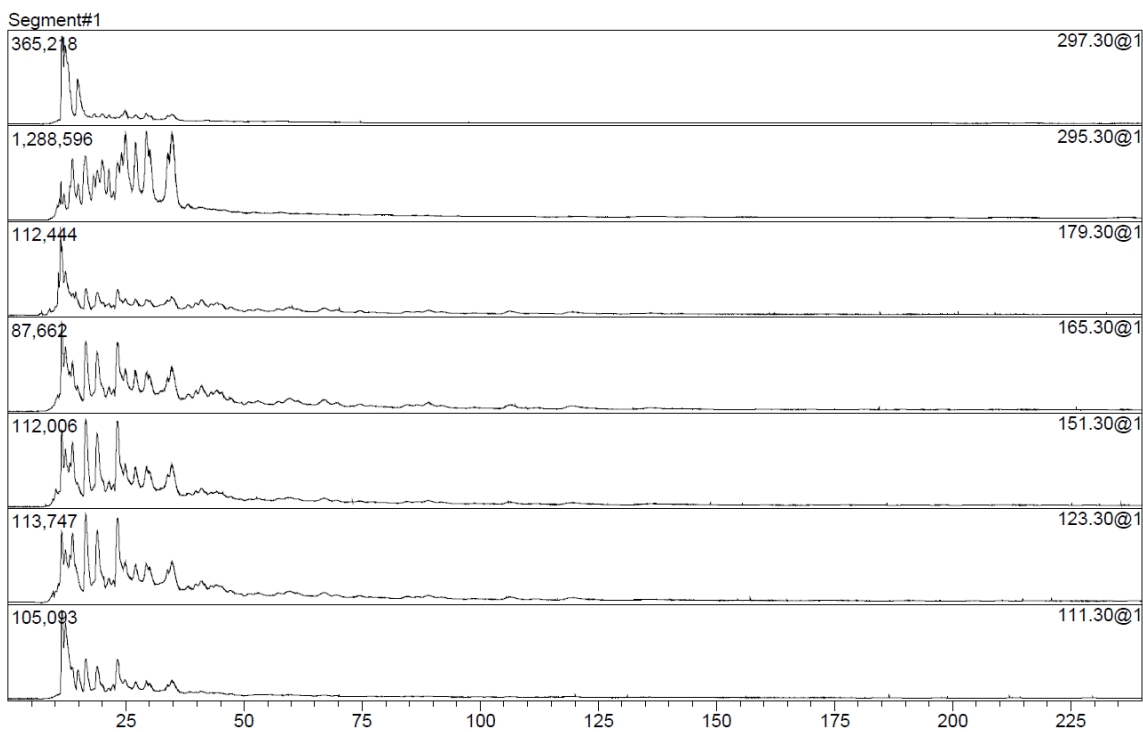


**Figure S16.** HPLC chiral purity of R and S isomers of archaeol **2** using chiral chromatography in reverse mode. Methanol:H<sub>2</sub>O 95:5% v/v was used as mobile phases at a flow of 0.4 mL/min with a post-column addition of 5 mM ammonium acetate at a flow of 0.1 mL/min.



**Figure S17.** Total ion and extracted ion ( $m/z$  653.65, 670.65 and 373.35) chromatograms for archaeol **3** (mixture of stereoisomers) using chiral chromatography in reverse mode. Methanol:H<sub>2</sub>O 95:5% v/v was used as mobile phases at a flow of 0.4 mL/min with a post-column addition of 5 mM ammonium acetate at a flow of 0.1 mL/min.

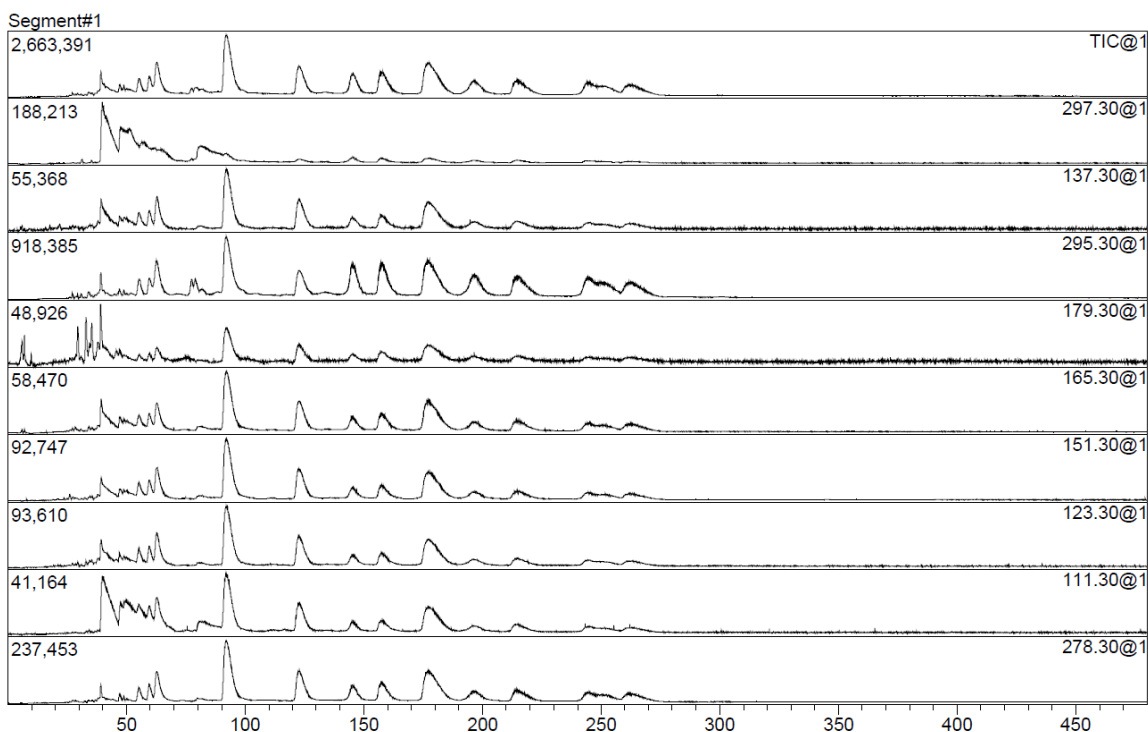




**Figure S18.** Total ion and extracted ion (molecular ion  $[M+H]^+$  at  $m/z$  297.3, reported fragments 295, 278, 165, 151, 123 and 111) chromatograms for phytol **7** using chiral chromatography in reverse mode. Methanol:H<sub>2</sub>O 95:5% v/v was used as mobile phases at a flow of 0.4 mL/min with a post-column addition of 5 mM ammonium acetate at a flow of 0.1 mL/min.

## 9. Chiral chromatography (reverse phase) for elucidation of stereoisomers

An effort was made to have better separation between peaks. Phytol **7** (97% mixture of isomers) was analyzed and the molecular ion  $[M+H]^+$  at  $m/z$  297.3152, reported fragments 295, 278, 165, 151, 123 and 111, were also monitored. A single quadrupole mass spectrometer was used for detection and was set to Selected Ion Mass (SIM) and operated in positive ionization mode. The chromatograms showed more than 15 peaks for the same molecular ion  $[M+H]^+$  at  $m/z$  297.30 and reported fragments. This correlates with the highly complex chiral structure observed in the chiral chromatography of optically inactive archaeol **3**. Better peak separation was achieved using this method.



**Figure S19.** Total ion and extracted ion (Molecular ion  $[M+H]^+$  at  $m/z$  297.30, reported fragments 295, 278, 165, 151, 123 and 111) chromatograms for phytol **7** using chiral chromatography in reverse mode. Methanol:H<sub>2</sub>O 85:15% v/v was used as mobile phases at a flow of 0.4 mL/min with a post-column addition of 5 mM ammonium acetate at a flow of 0.1 mL/min.

## 10. Percentage of archaeol 3 in 100% R configuration

<Chromatogram>

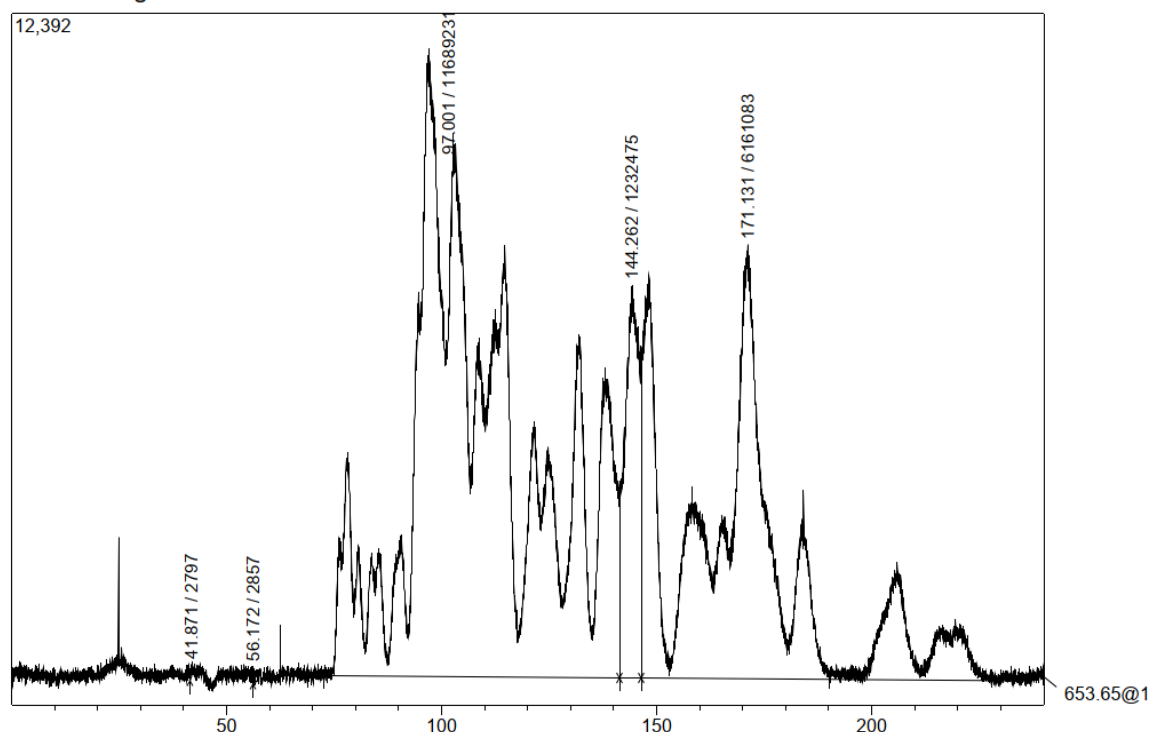


Figure S20: Chromatogram obtained from the chiral chromatography of optically inactive archaeol **3**, displaying a highly complex chiral composition with more than 20 peaks for the same molecular ion  $[M+H]^+$  at  $m/z$  653.65.

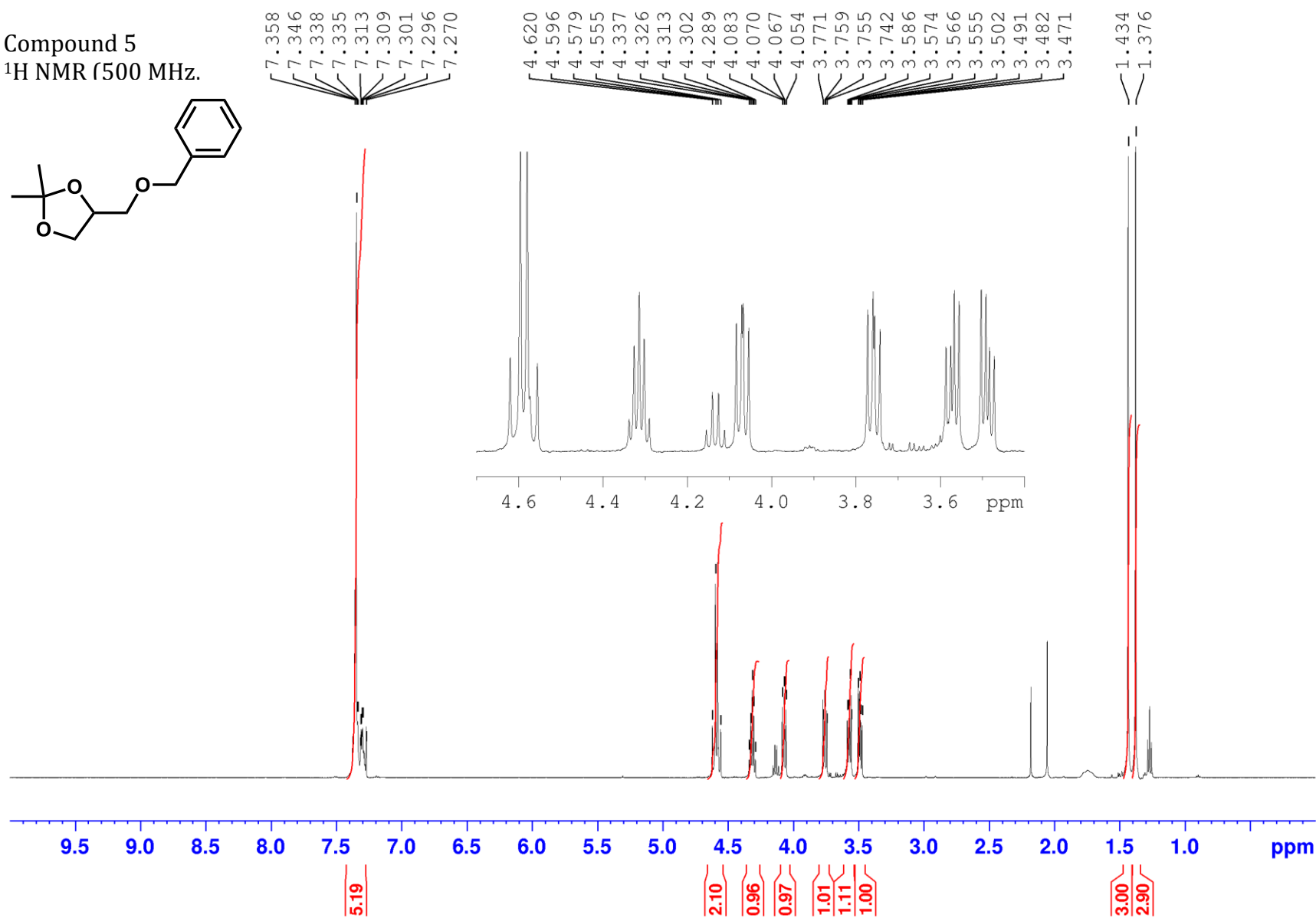
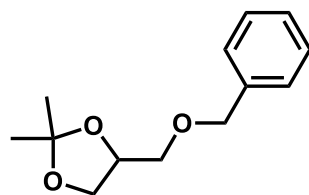
**Table S2. Calculation of the percentage of archaeol 3 in the 100% R configuration**

Ion $m/z$	Total Peak Area before R	Total Peak Area after R	Peak Area (R) RT: 144.2 min	Total Area	% of (R)-2,3-bis(((3R,7R,11R)-3,7,11,15-tetramethylhexadecyl)oxy)propan-1-ol (Archaeol <b>1</b> )
653.65	11689231	6161083	1232475	19082789	6.46%

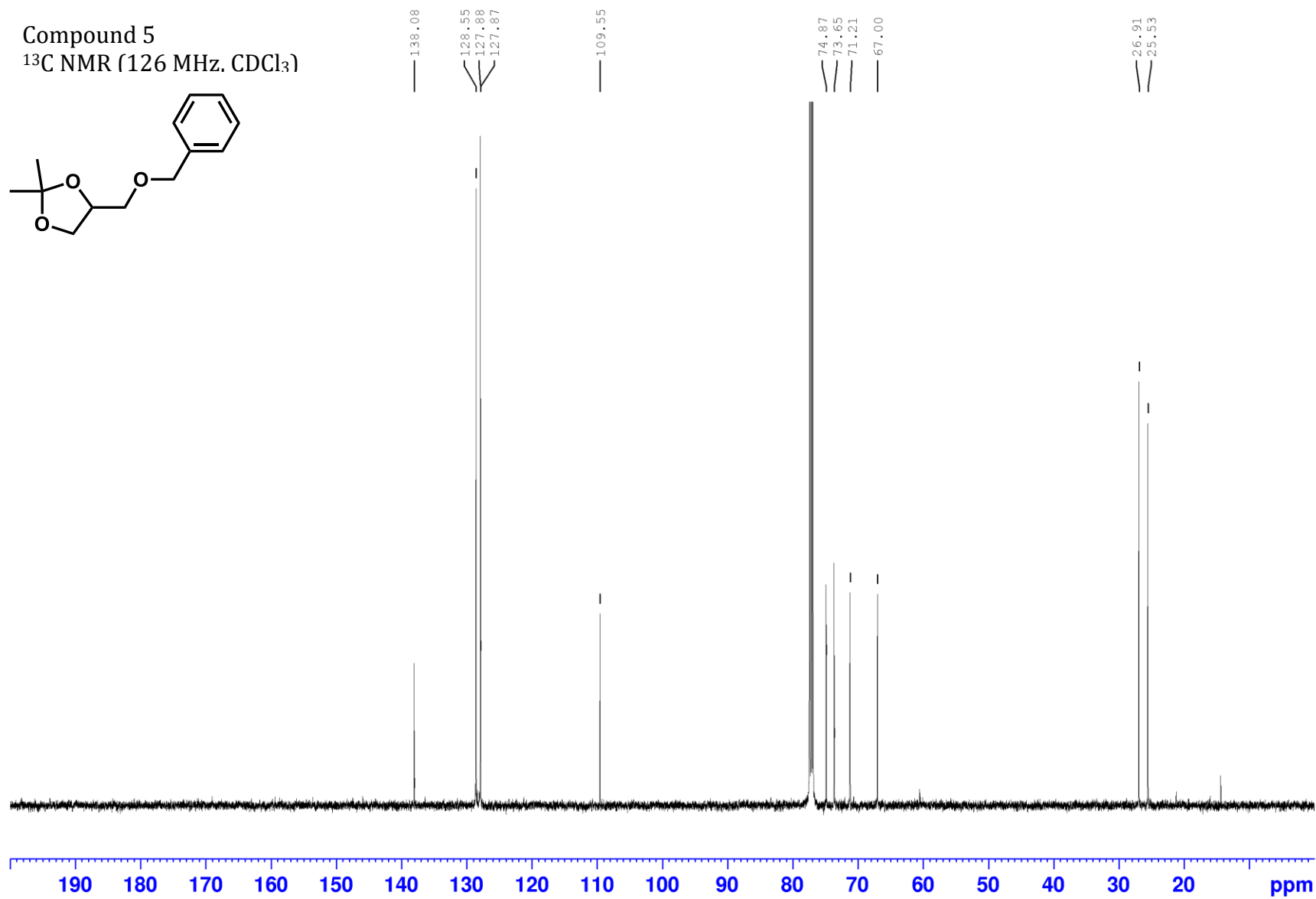
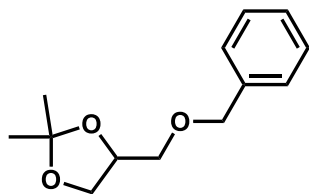
This chromatogram was used to estimate the percentage of archaeol **3** corresponding to the single stereoisomer of archaeol **1**. It was determined by dividing the peak area of archaeol **3** at 144.2 minutes by the total area for all peaks (Table S2). Approximately 6.5% of archaeol **3** was found to have the 100% R configuration. It can therefore be extrapolated that while 100% of SLA contains 100% of archaeol **3**, 100% of **SLA-3** only contains about

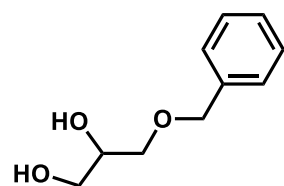
6.5% of archaeol **1**. The remaining 93.5% of **SLA-3** contains other stereoisomers of archaeol.

Compound 5  
 $^1\text{H}$  NMR (500 MHz,

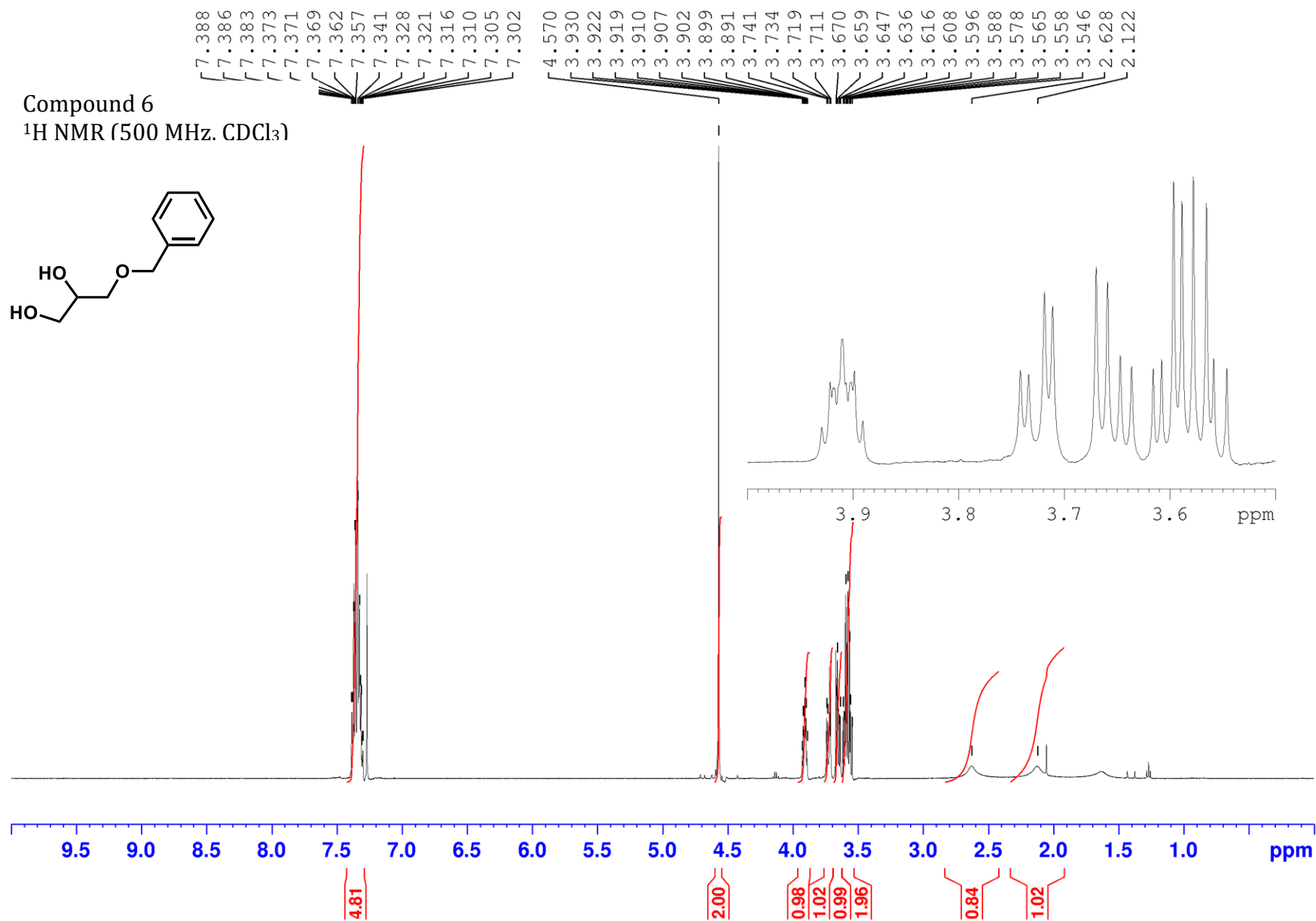


Compound 5  
 $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$ )

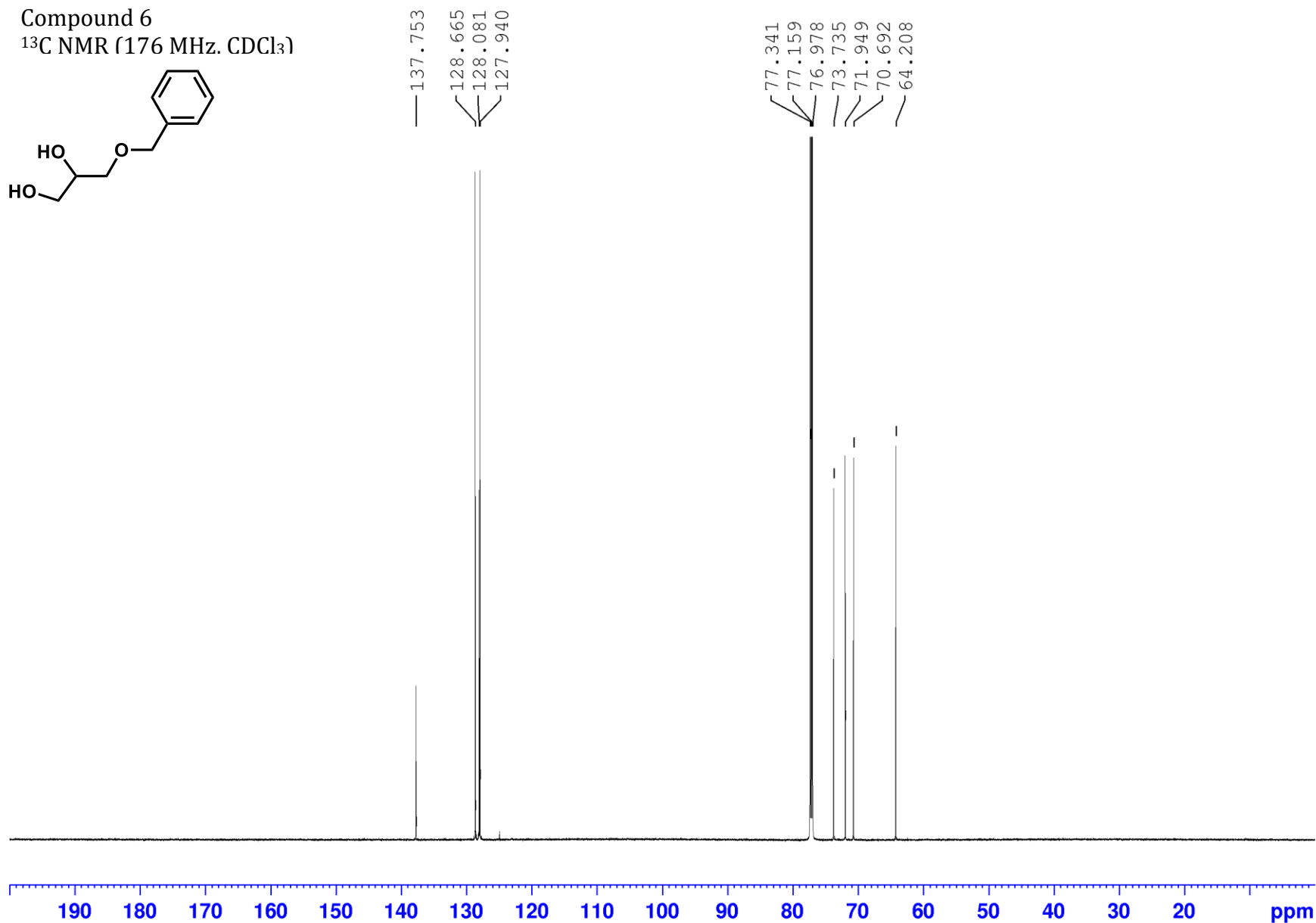
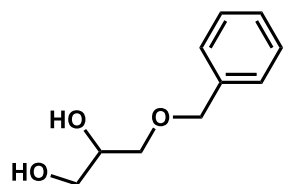




Compound 6  
 $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )

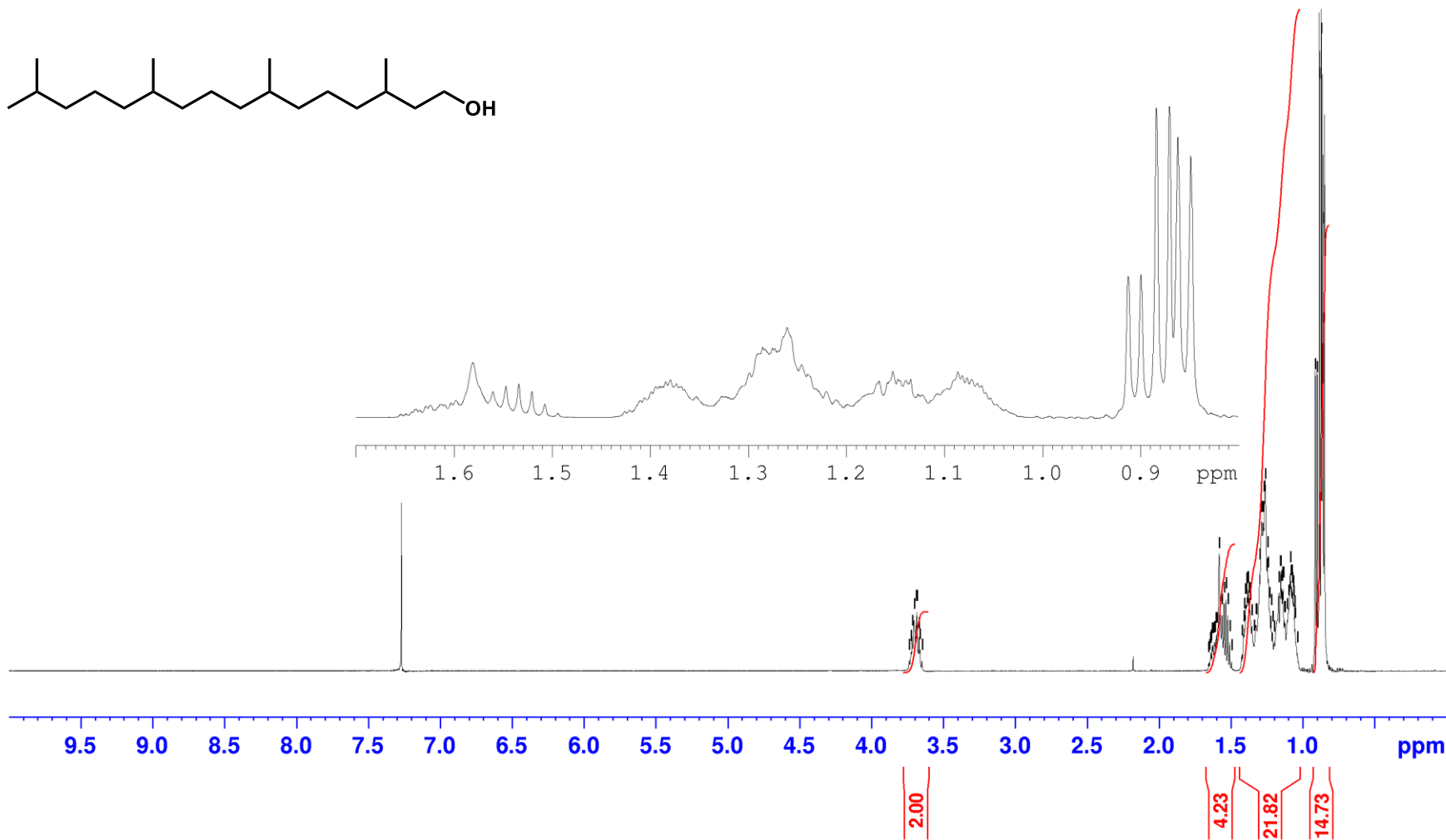
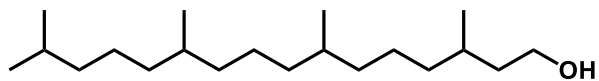


Compound 6  
 $^{13}\text{C}$  NMR (176 MHz,  $\text{CDCl}_3$ )



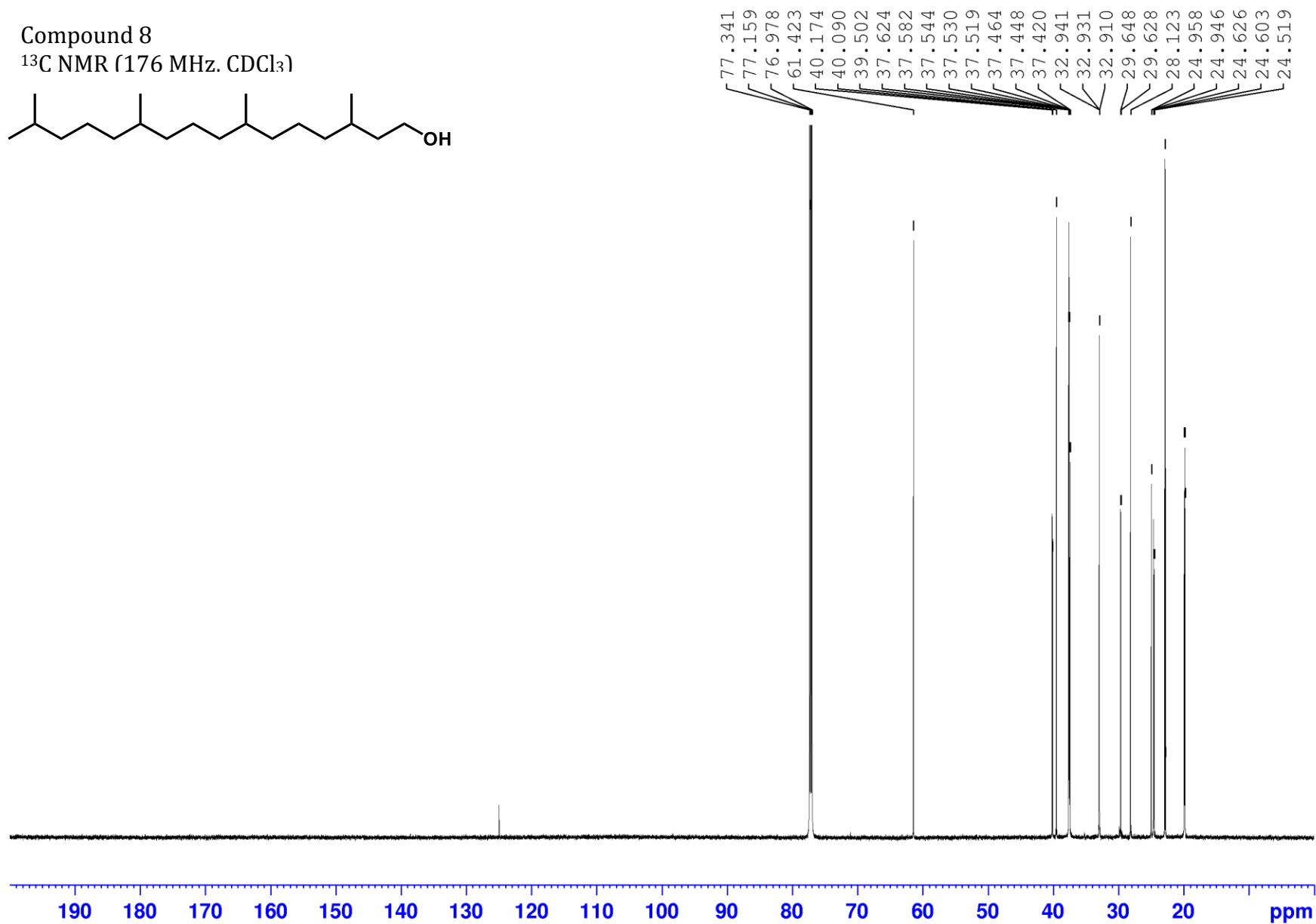
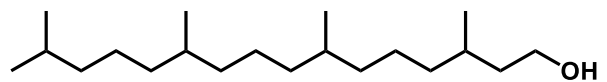


3.702  
 3.689  
 3.684  
 1.581  
 1.560  
 1.547  
 1.534  
 1.521  
 1.405  
 1.399  
 1.394  
 1.390  
 1.388  
 1.384  
 1.379  
 1.373  
 1.368  
 1.365  
 1.353  
 1.327  
 1.323  
 1.299  
 1.289  
 1.285  
 1.282  
 1.275  
 1.272  
 1.269  
 1.260  
 1.245  
 1.239  
 1.231  
 1.219  
 1.166  
 1.158  
 1.152  
 1.147  
 1.145  
 1.139  
 1.134  
 1.126  
 1.123  
 1.107  
 1.097  
 1.086  
 1.081  
 1.076  
 1.072  
 1.066  
 1.061  
 1.057  
 1.053  
 0.912  
 0.899  
 0.883  
 0.870  
 0.861  
 0.848

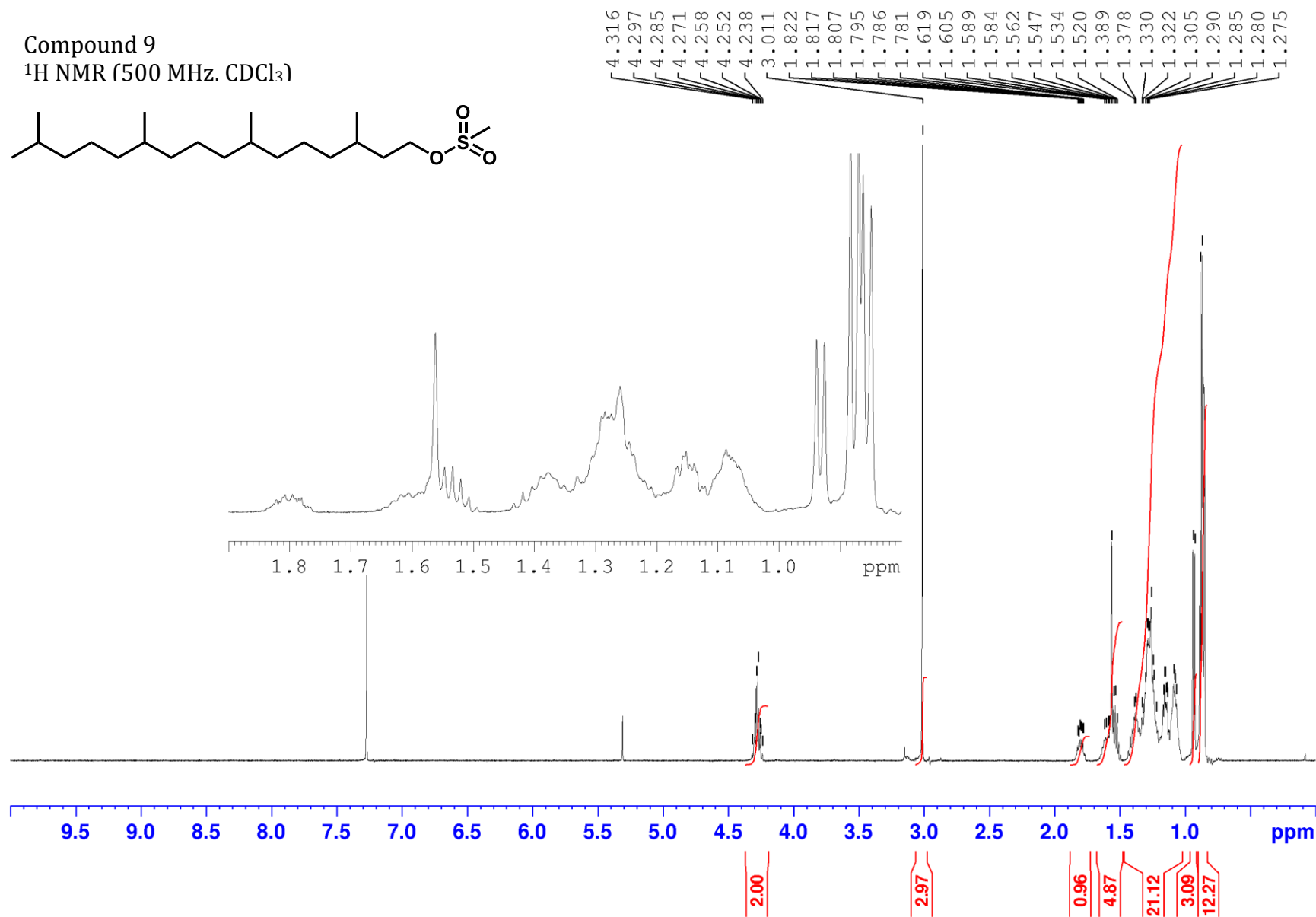
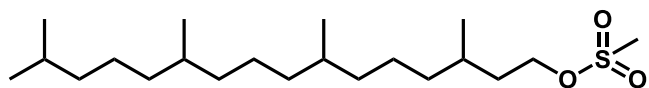


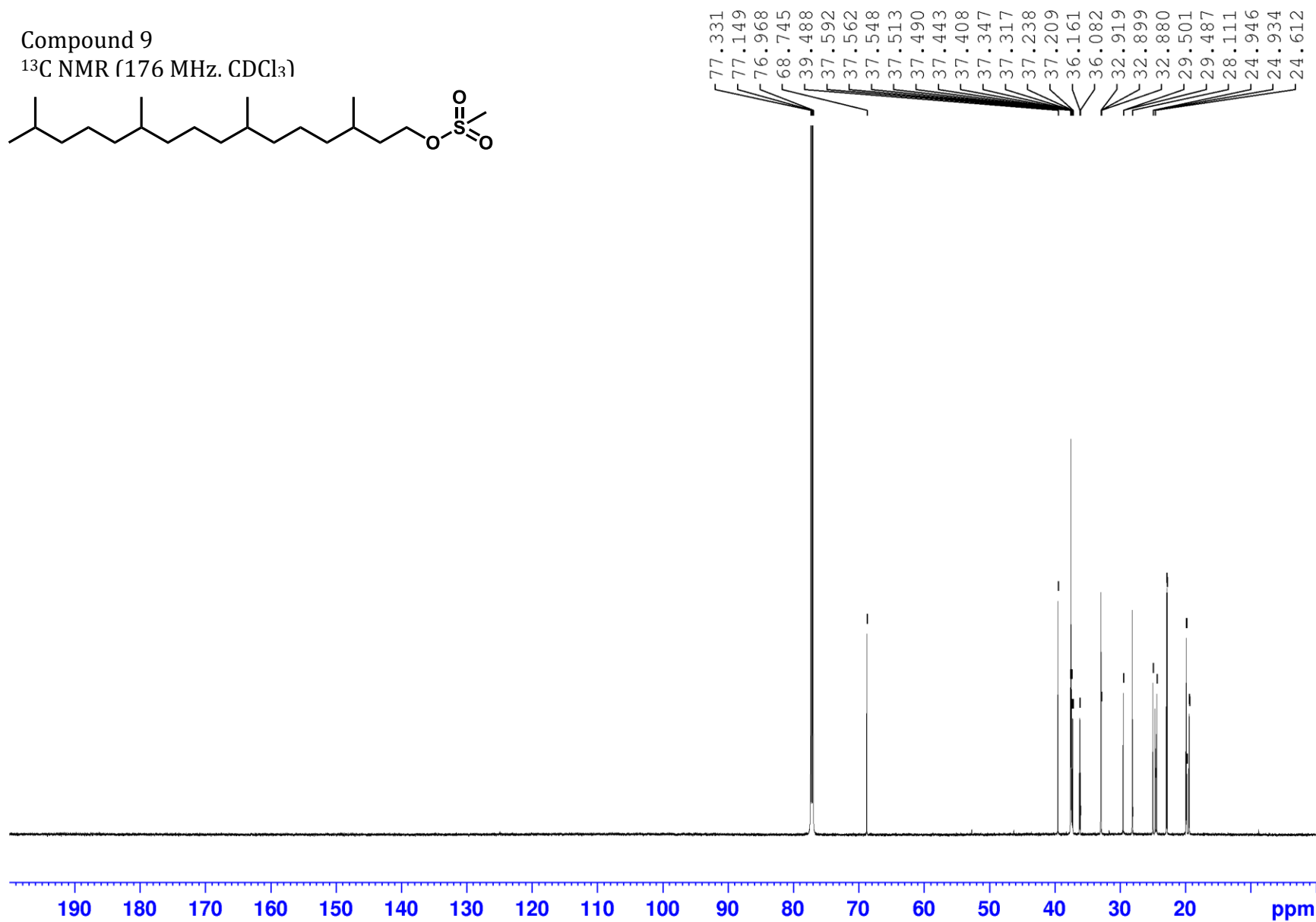
Compound 8

$^{13}\text{C}$  NMR (176 MHz,  $\text{CDCl}_3$ )

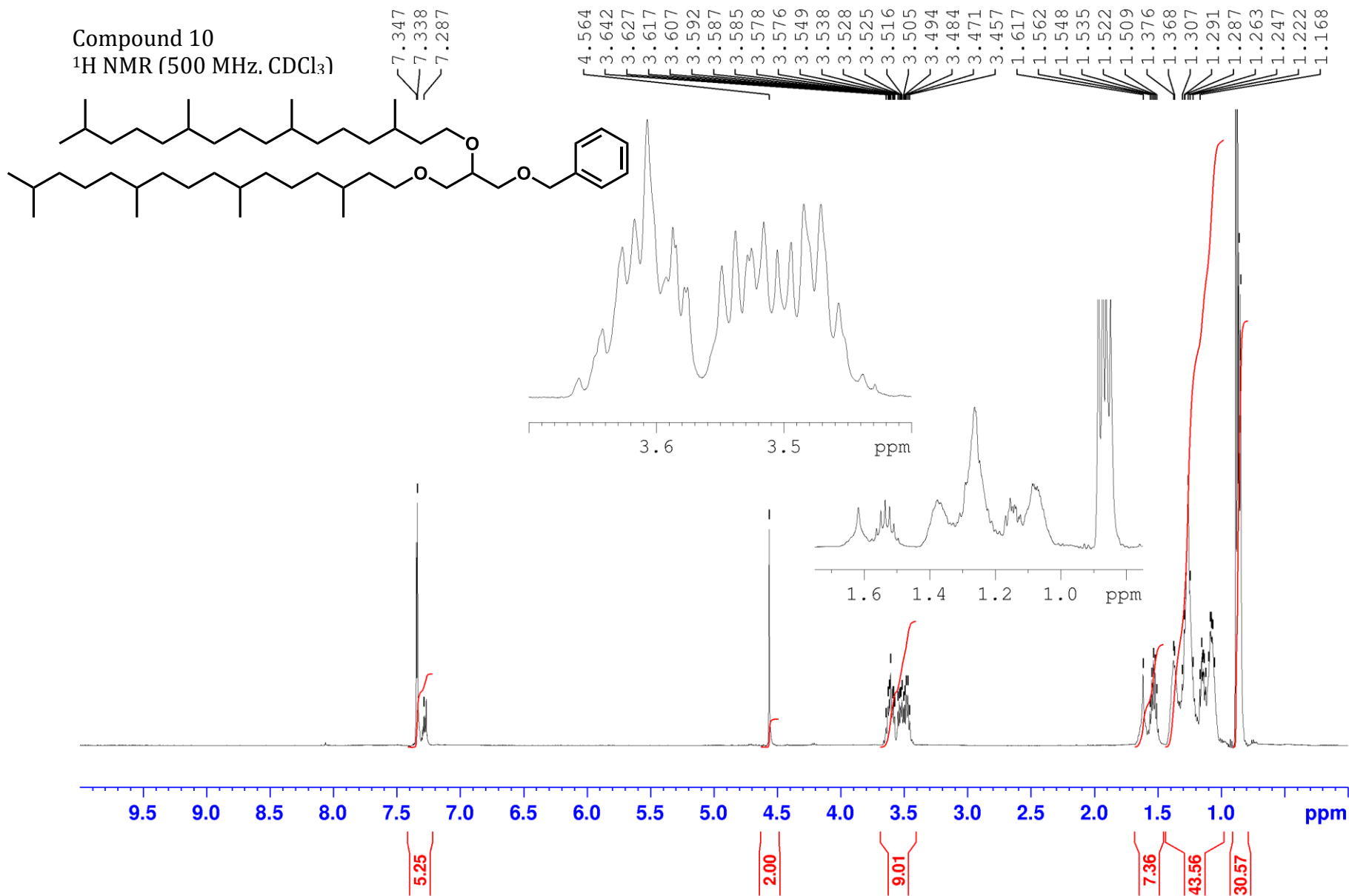


Compound 9  
 $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )

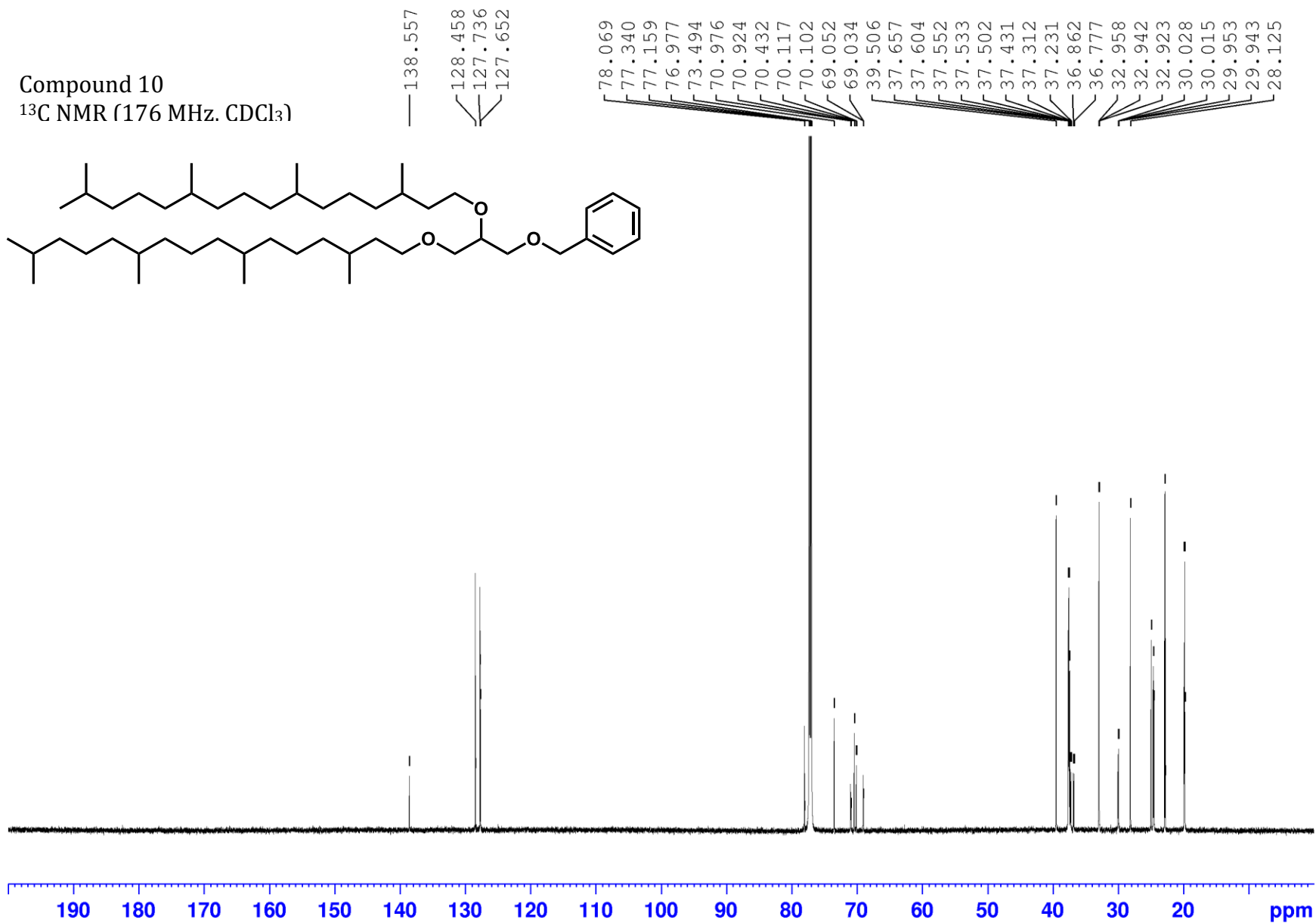


CC(C)CCCC(C)CCCC(C)CCCC(C)CCCC(C)OS(=O)(=O)C

Compound 10  
 $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )



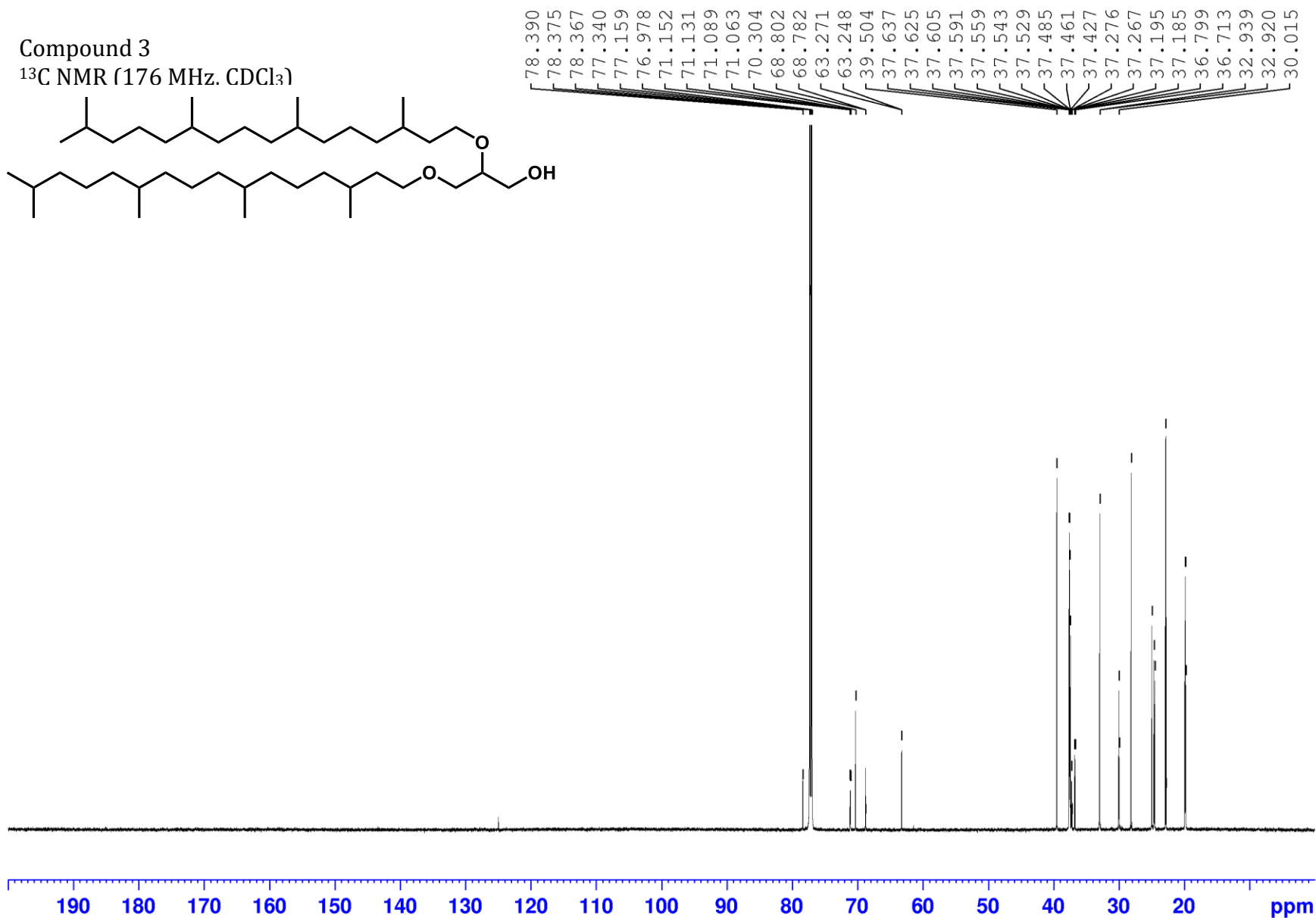
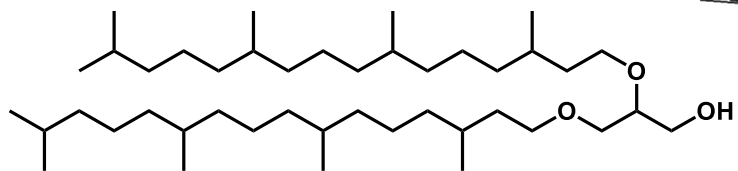
Compound 10  
 $^{13}\text{C}$  NMR (176 MHz,  $\text{CDCl}_3$ )



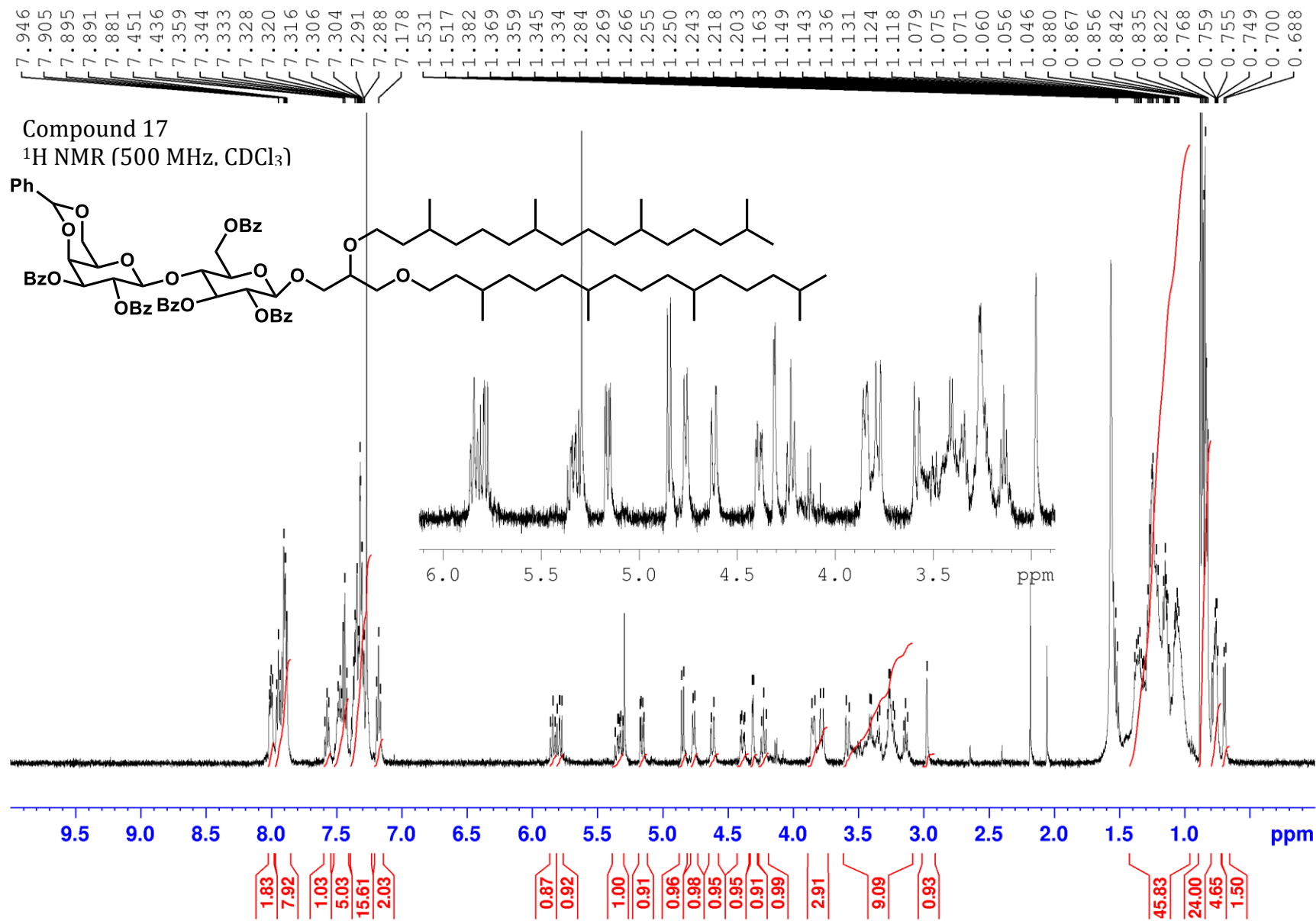


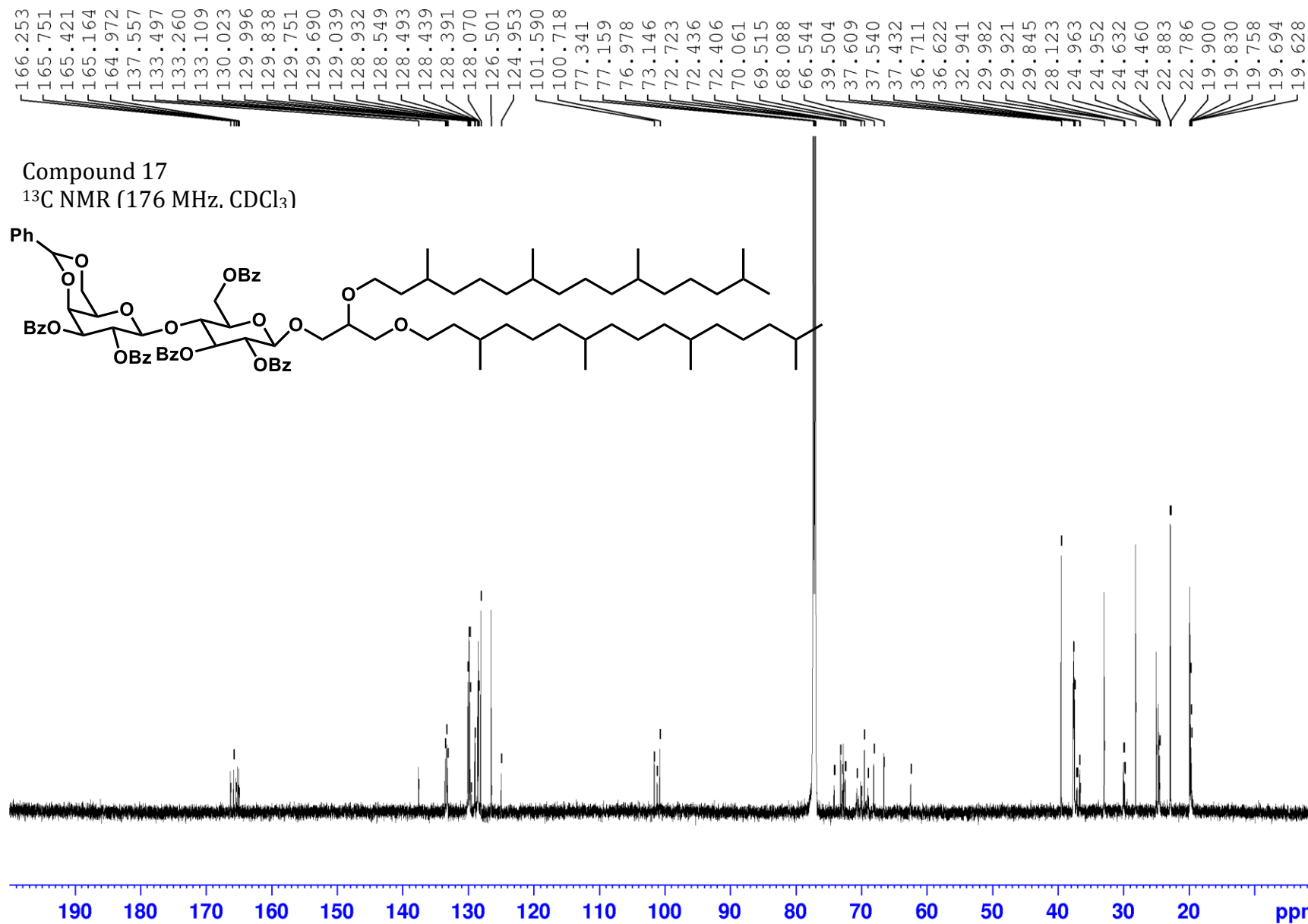
Compound 3

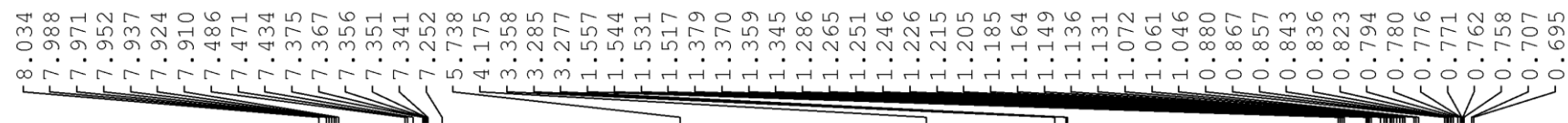
$^{13}\text{C}$  NMR (176 MHz,  $\text{CDCl}_3$ )



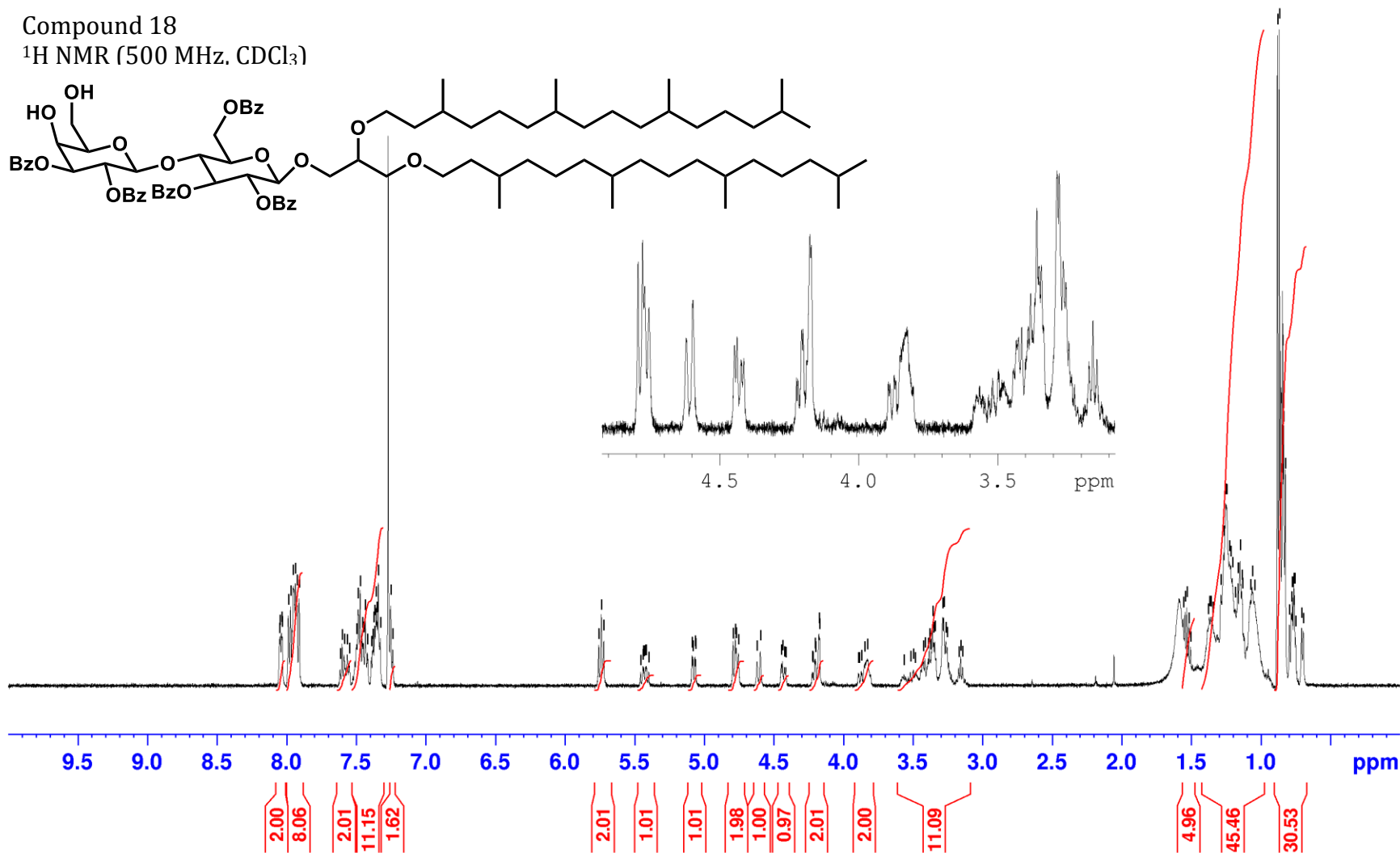
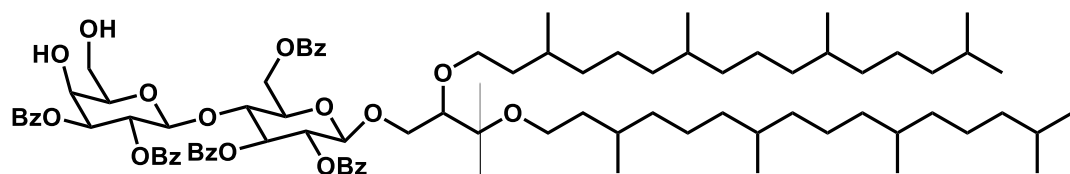


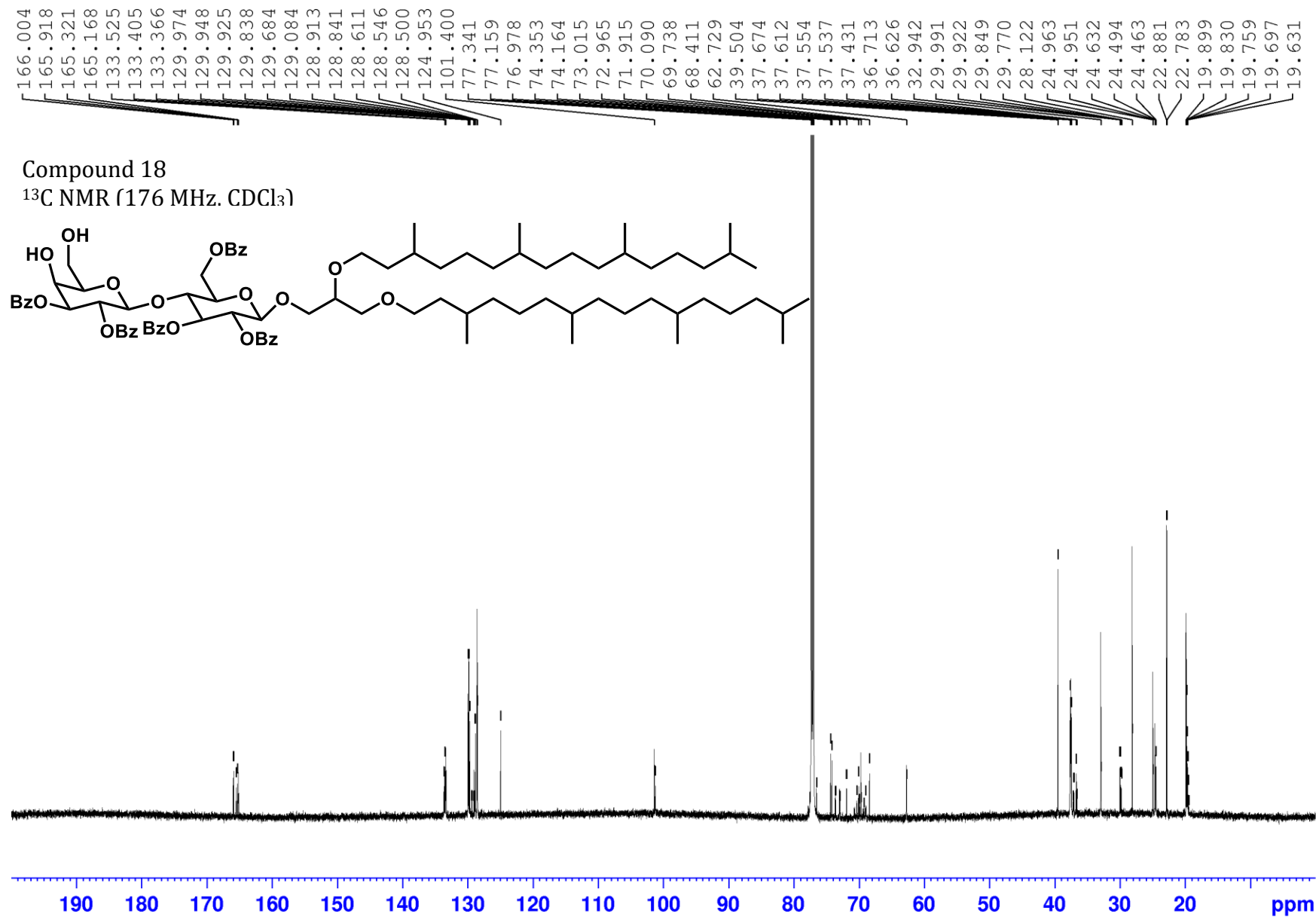


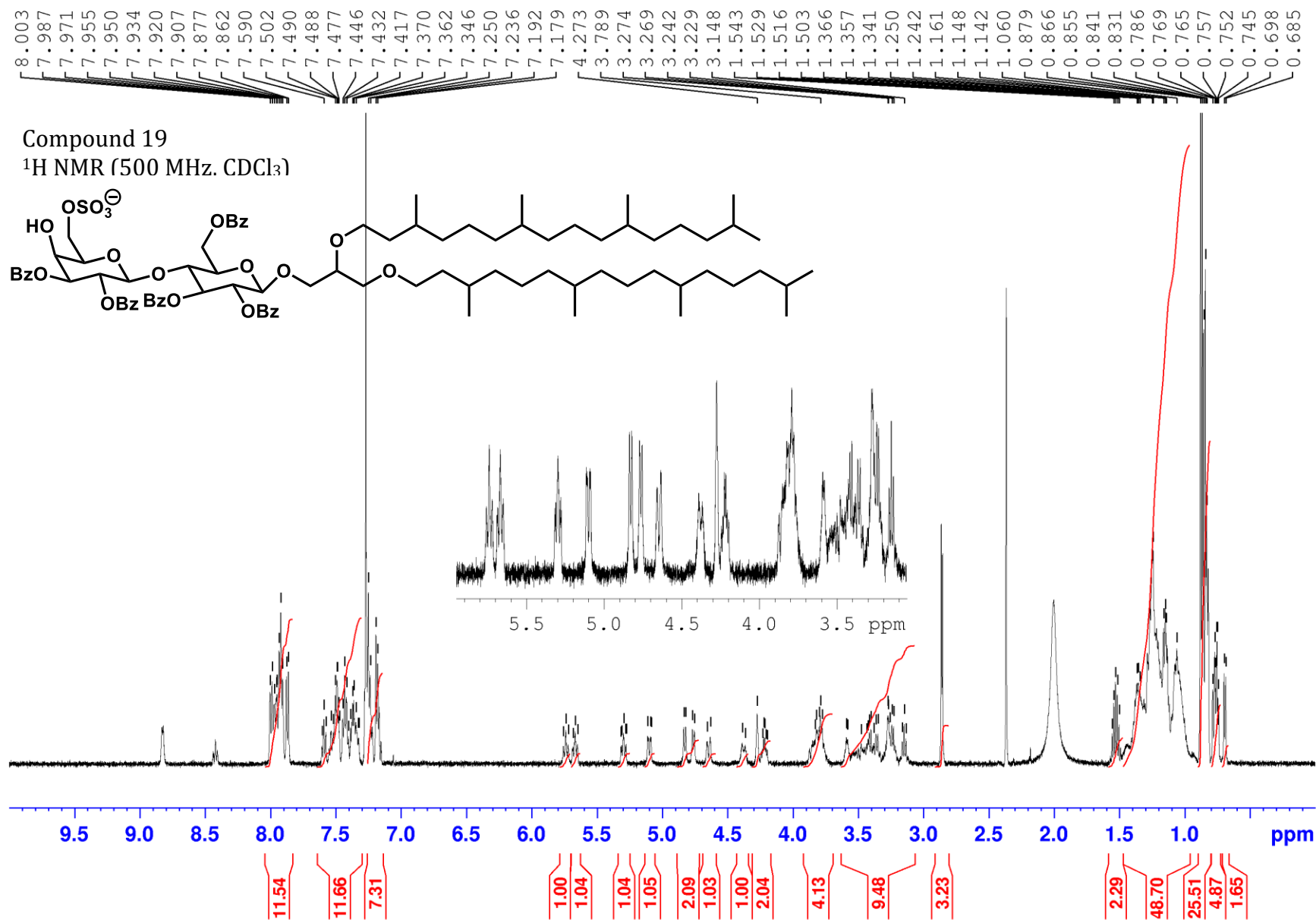


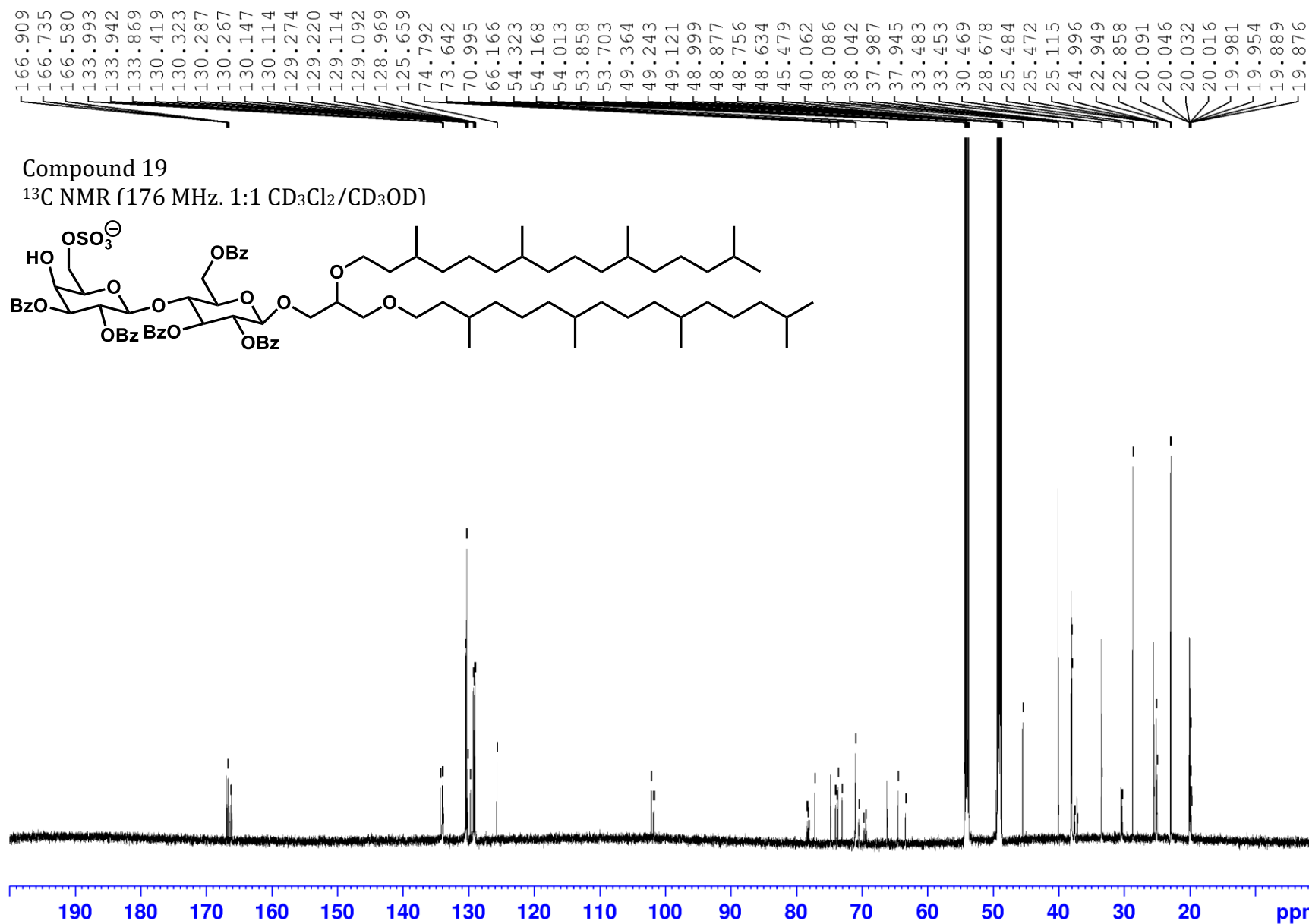


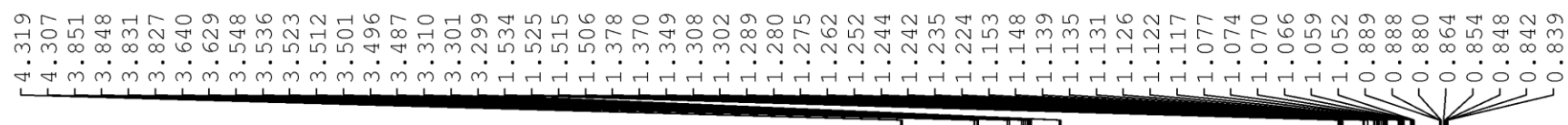
Compound 18  
<sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>)





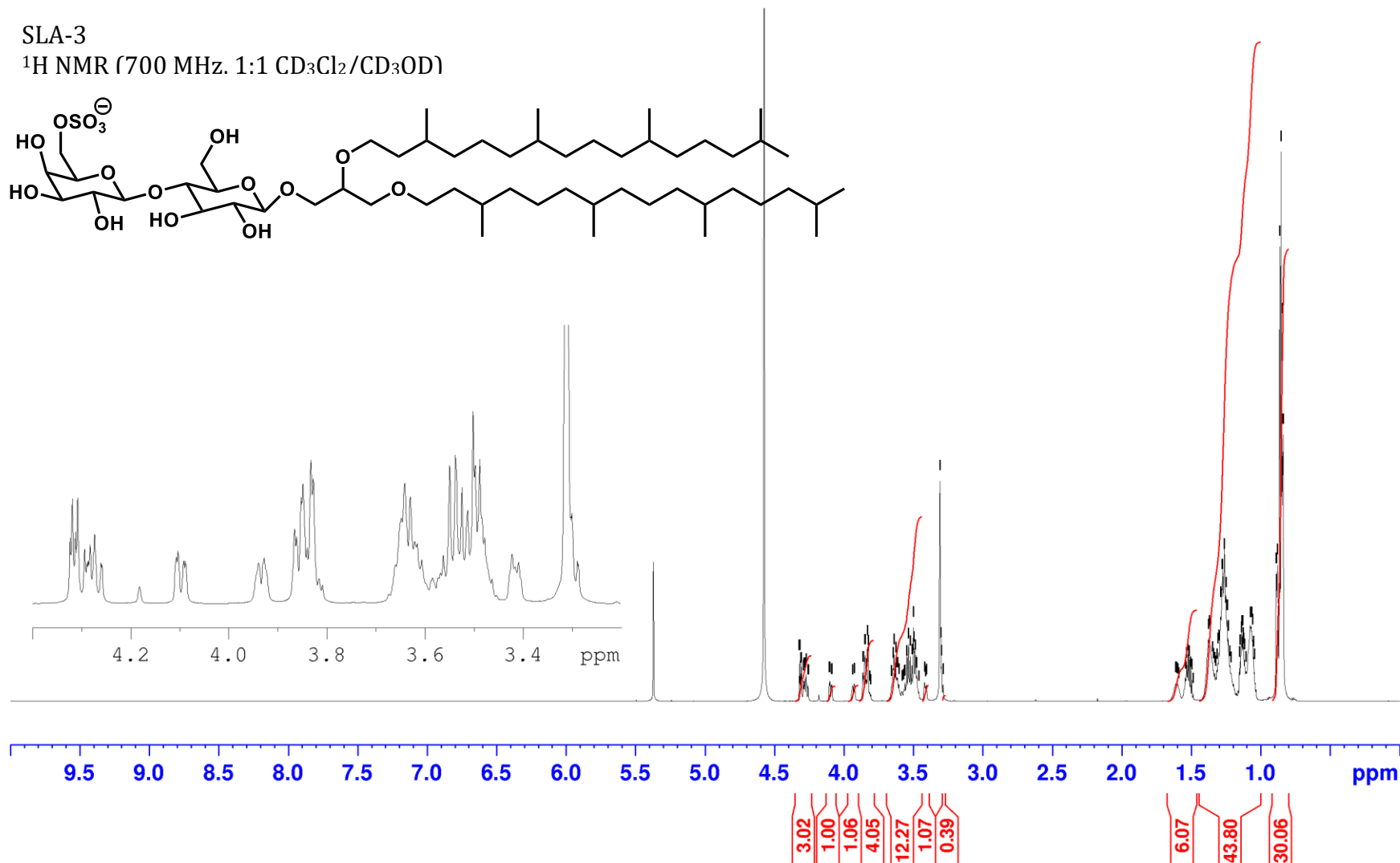
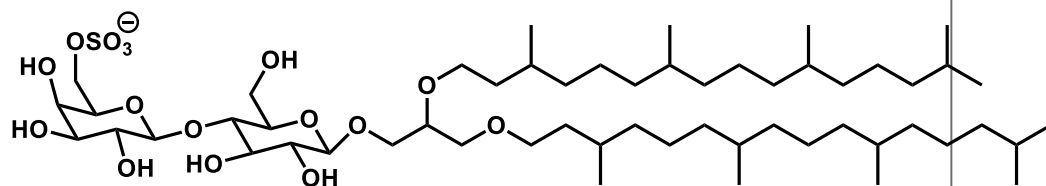


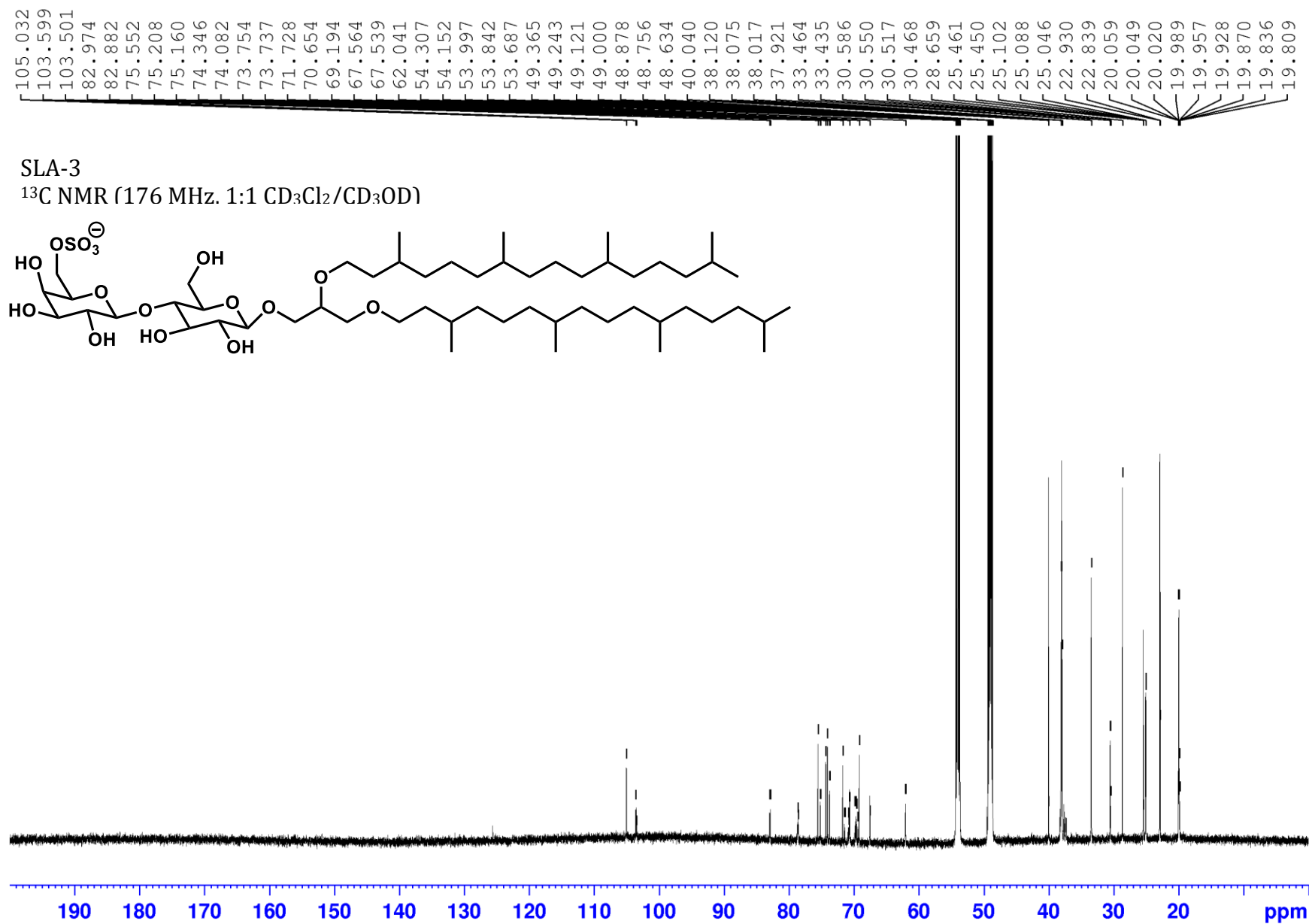




SLA-3

$^1\text{H}$  NMR (700 MHz, 1:1  $\text{CD}_3\text{Cl}_2/\text{CD}_3\text{OD}$ )



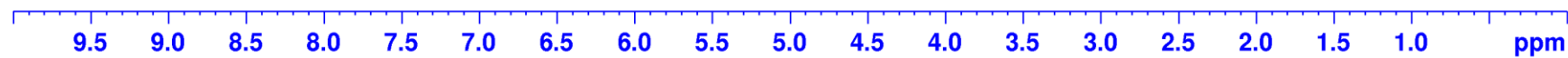
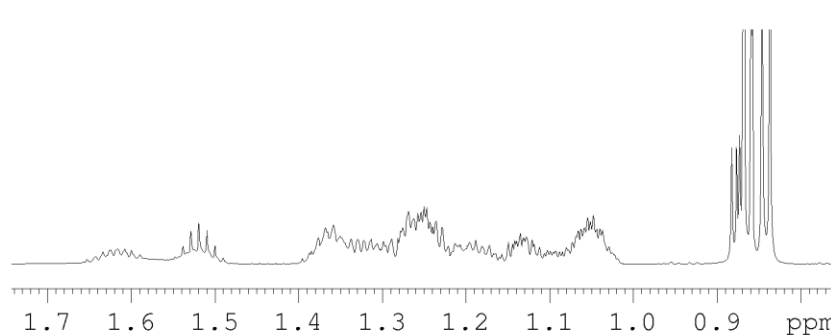
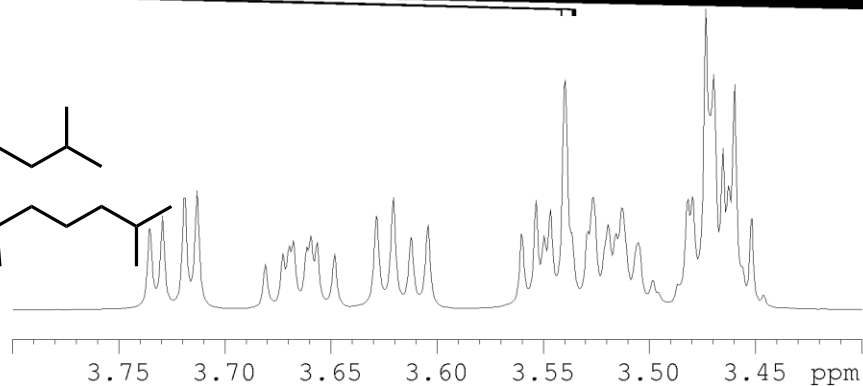
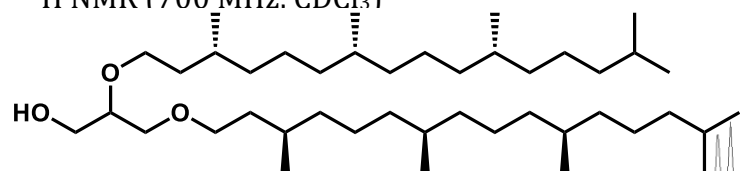




3.540  
3.473  
3.470  
3.460  
1.528  
1.519  
1.509  
1.376  
1.368  
1.365  
1.358  
1.350  
1.337  
1.329  
1.321  
1.313  
1.298  
1.288  
1.281  
1.278  
1.275  
1.268  
1.262  
1.257  
1.253  
1.250  
1.246  
1.243  
1.239  
1.235  
1.228  
1.188  
1.141  
1.139  
1.134  
1.131  
1.130  
1.128  
1.120  
1.068  
1.066  
1.062  
1.058  
1.054  
1.051  
1.047  
1.043  
1.039  
1.036  
0.882  
0.876  
0.872  
0.868  
0.867  
0.859  
0.857  
0.846  
0.836

Archaeol 2

$^1\text{H}$  NMR (700 MHz,  $\text{CDCl}_3$ )

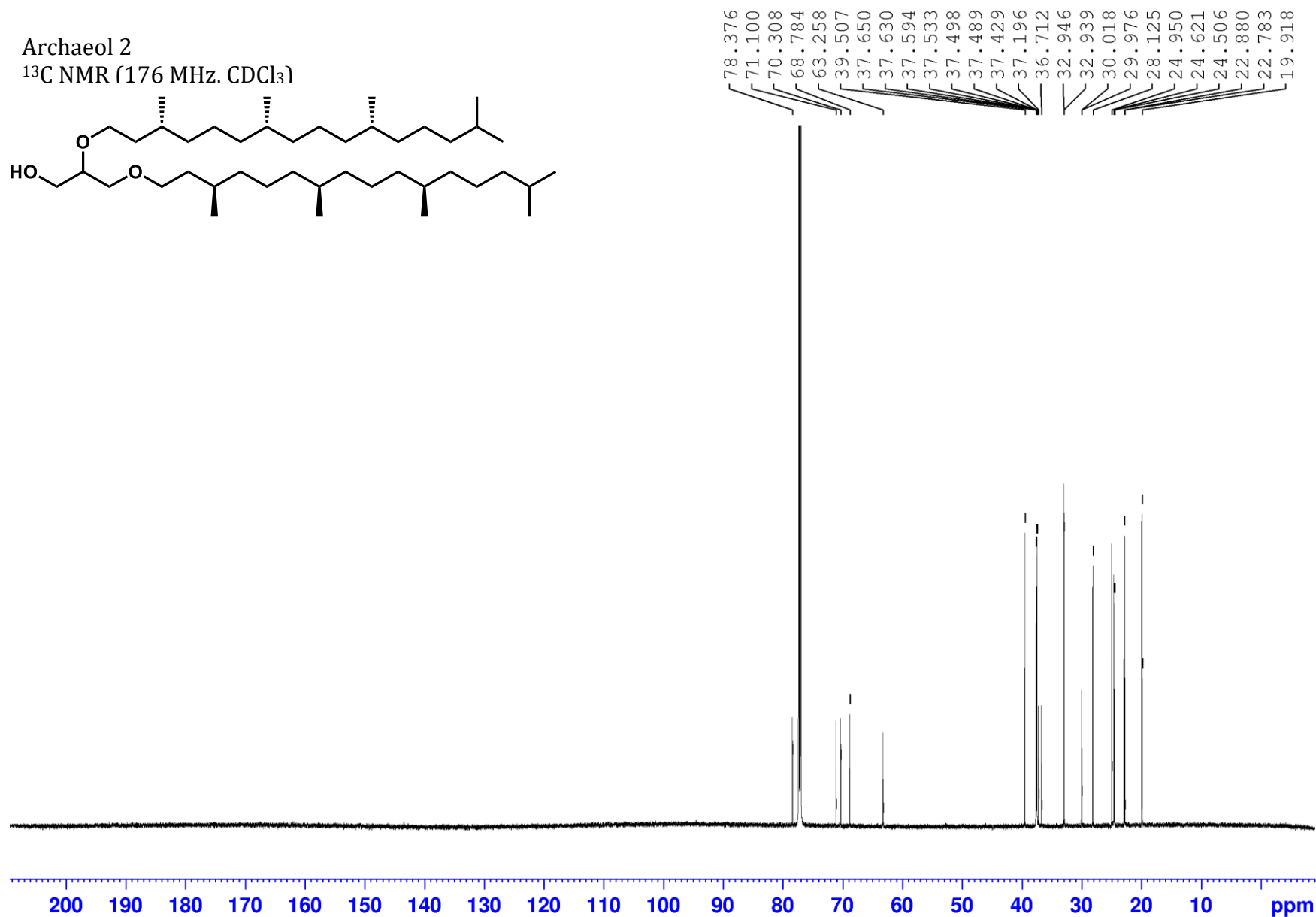
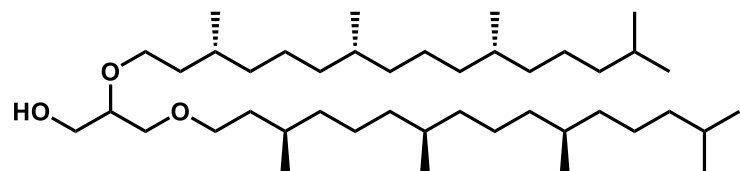


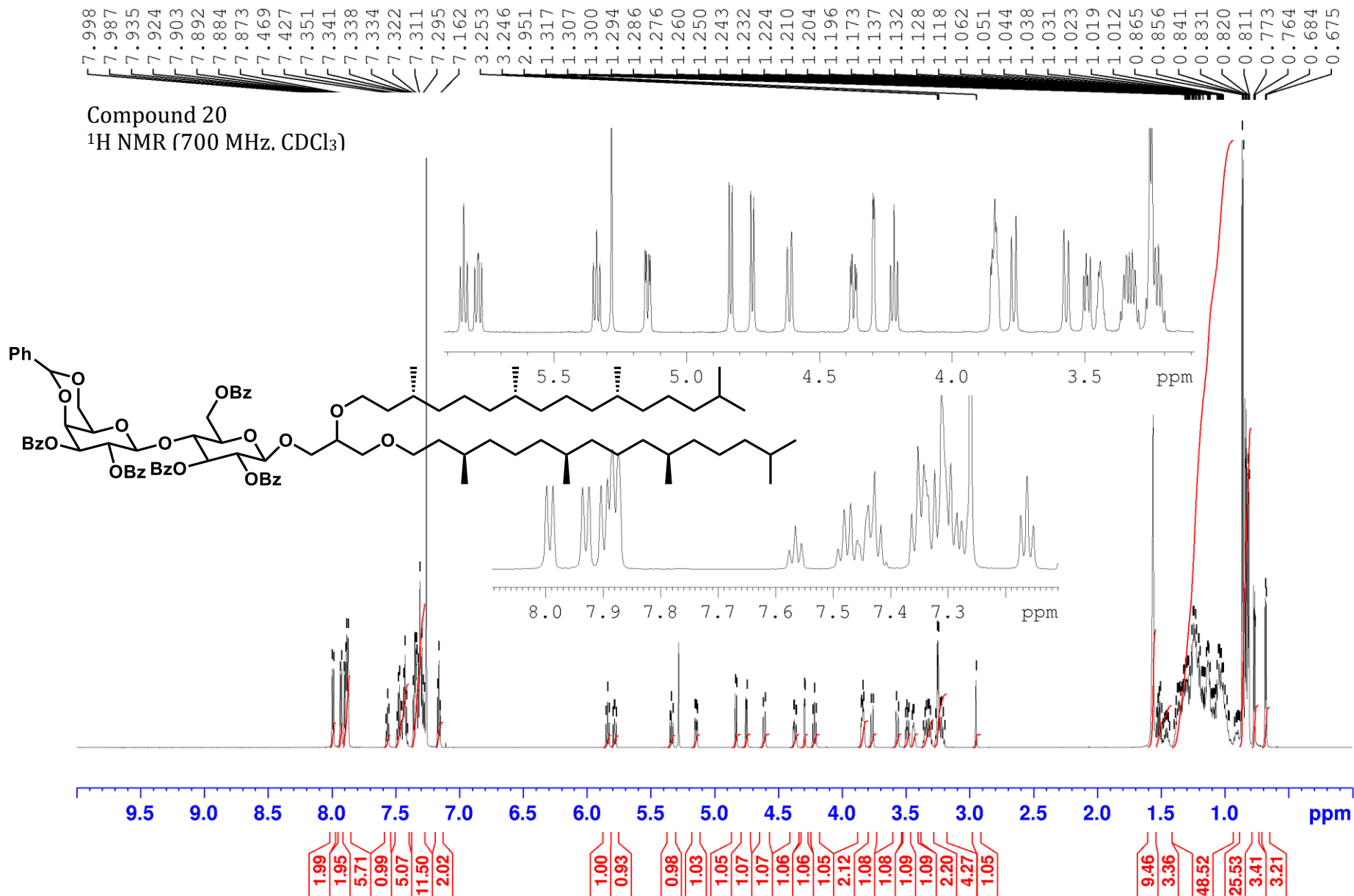
1.00  
0.99  
0.92  
2.88  
3.00

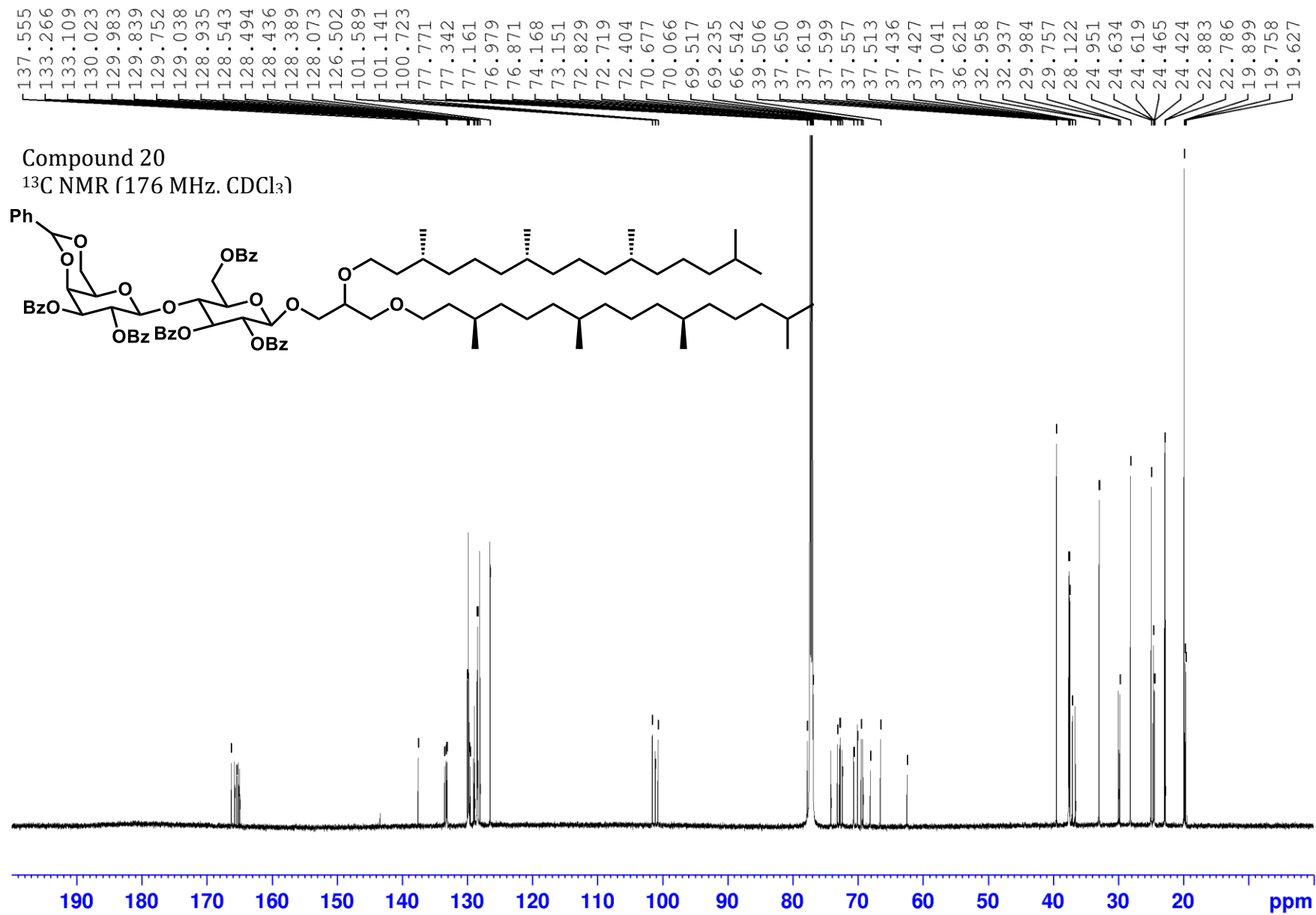
2.10  
3.54  
12.46  
8.33  
2.44  
3.16  
7.36  
17.60  
11.27

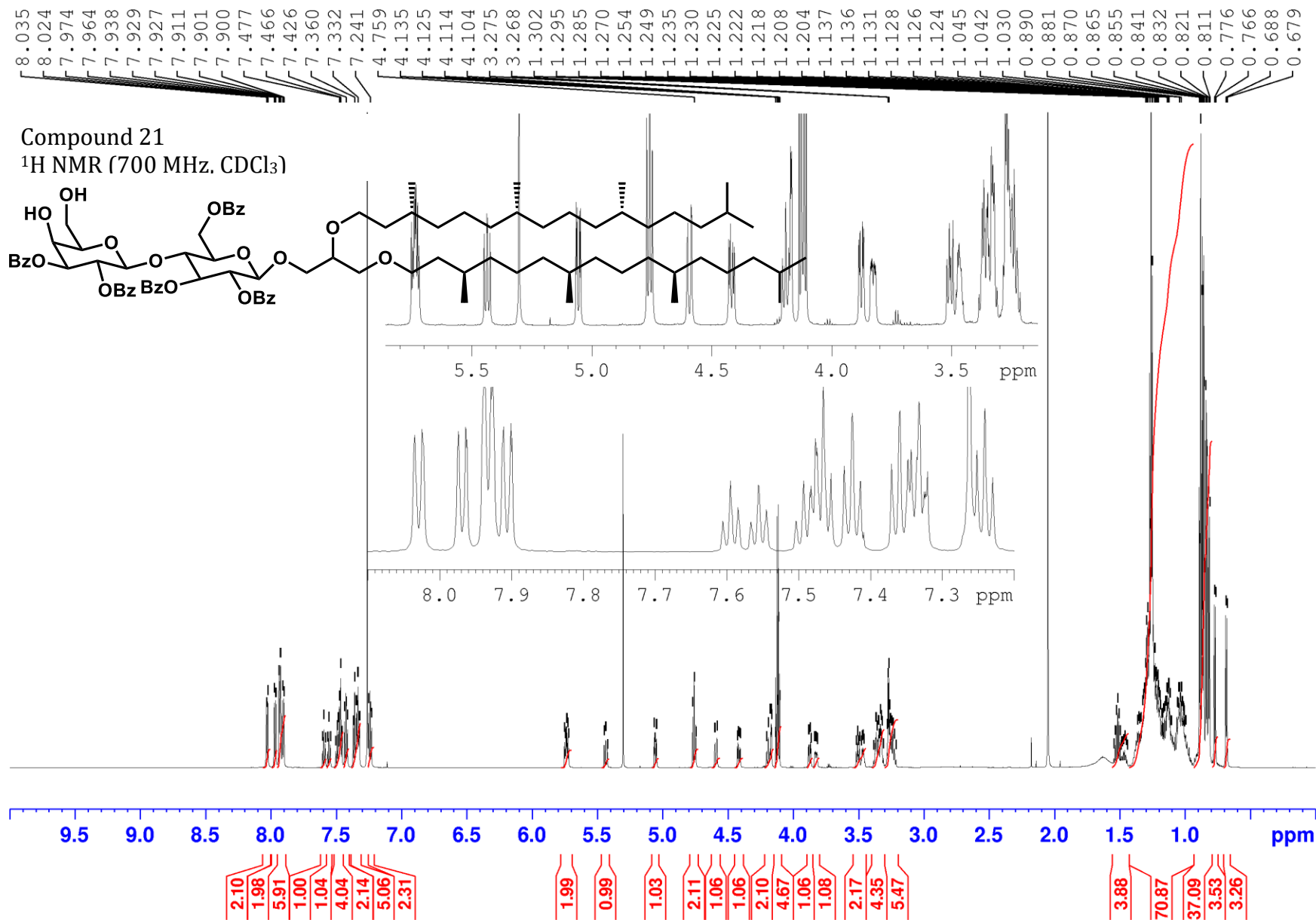
# Archaeol 2

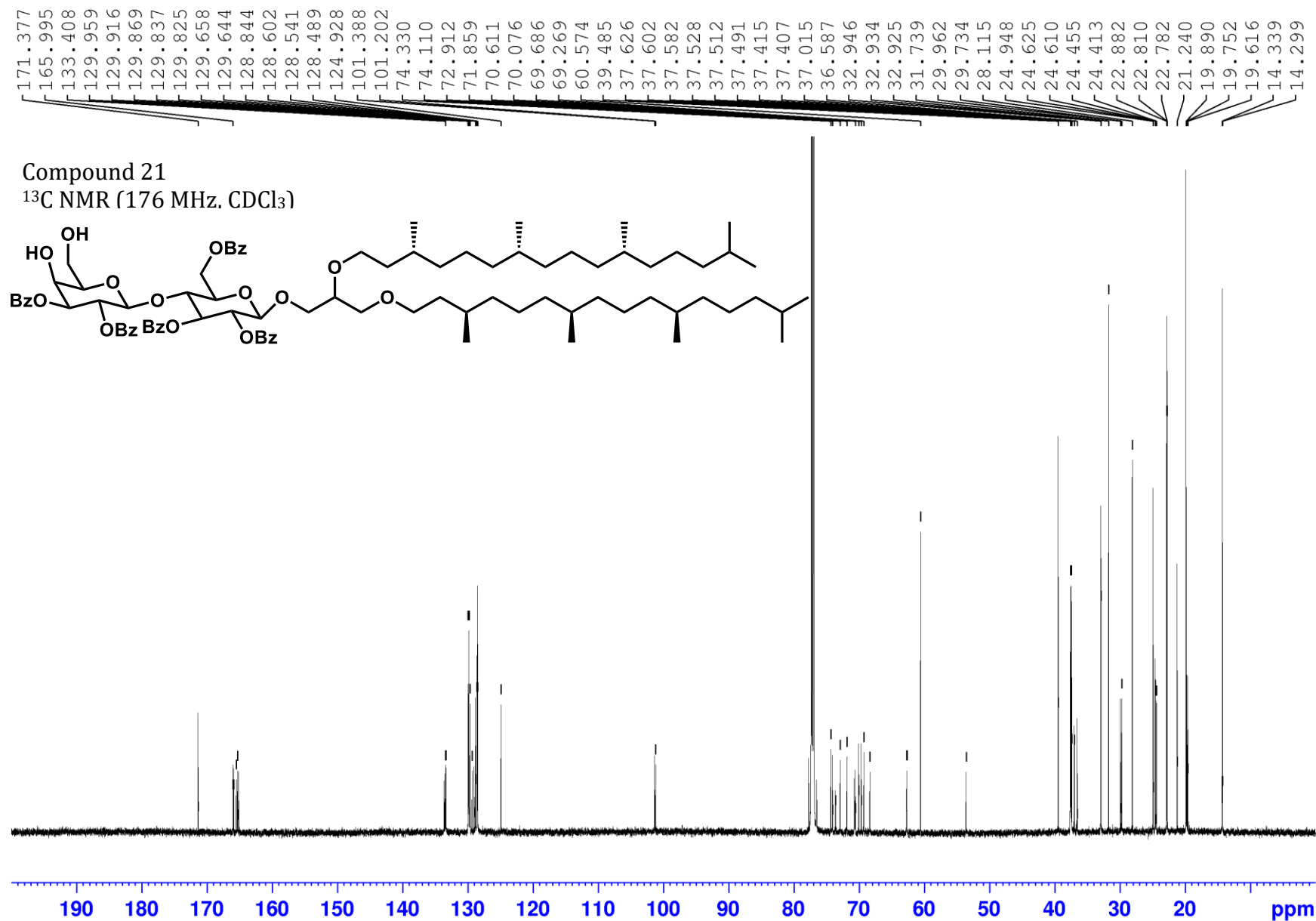
$^{13}\text{C}$  NMR (176 MHz,  $\text{CDCl}_3$ )

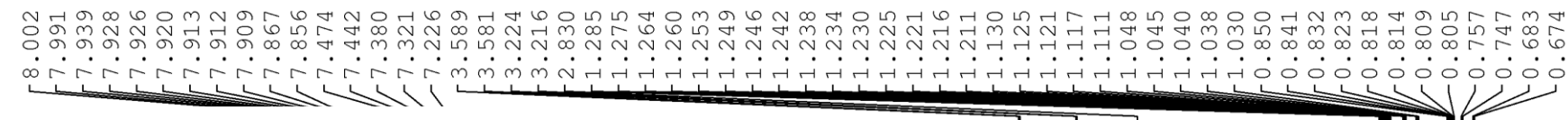






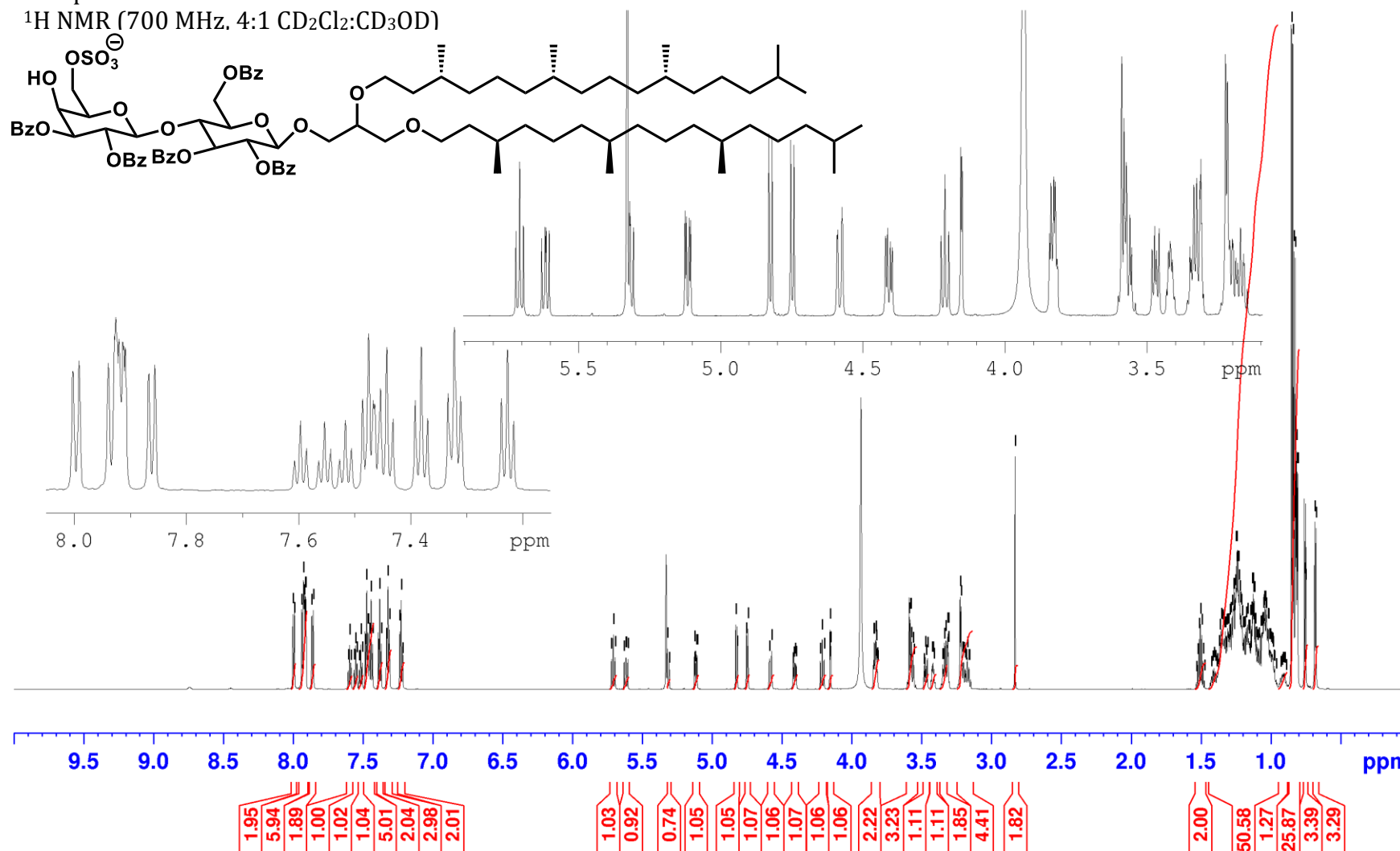
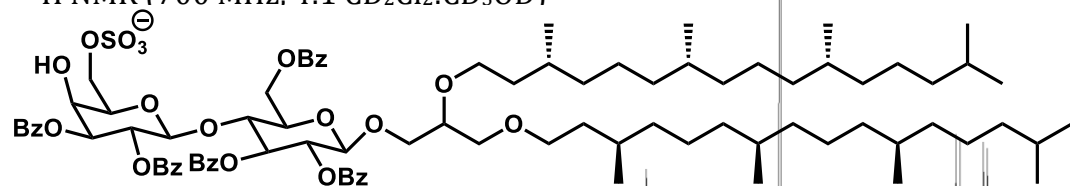


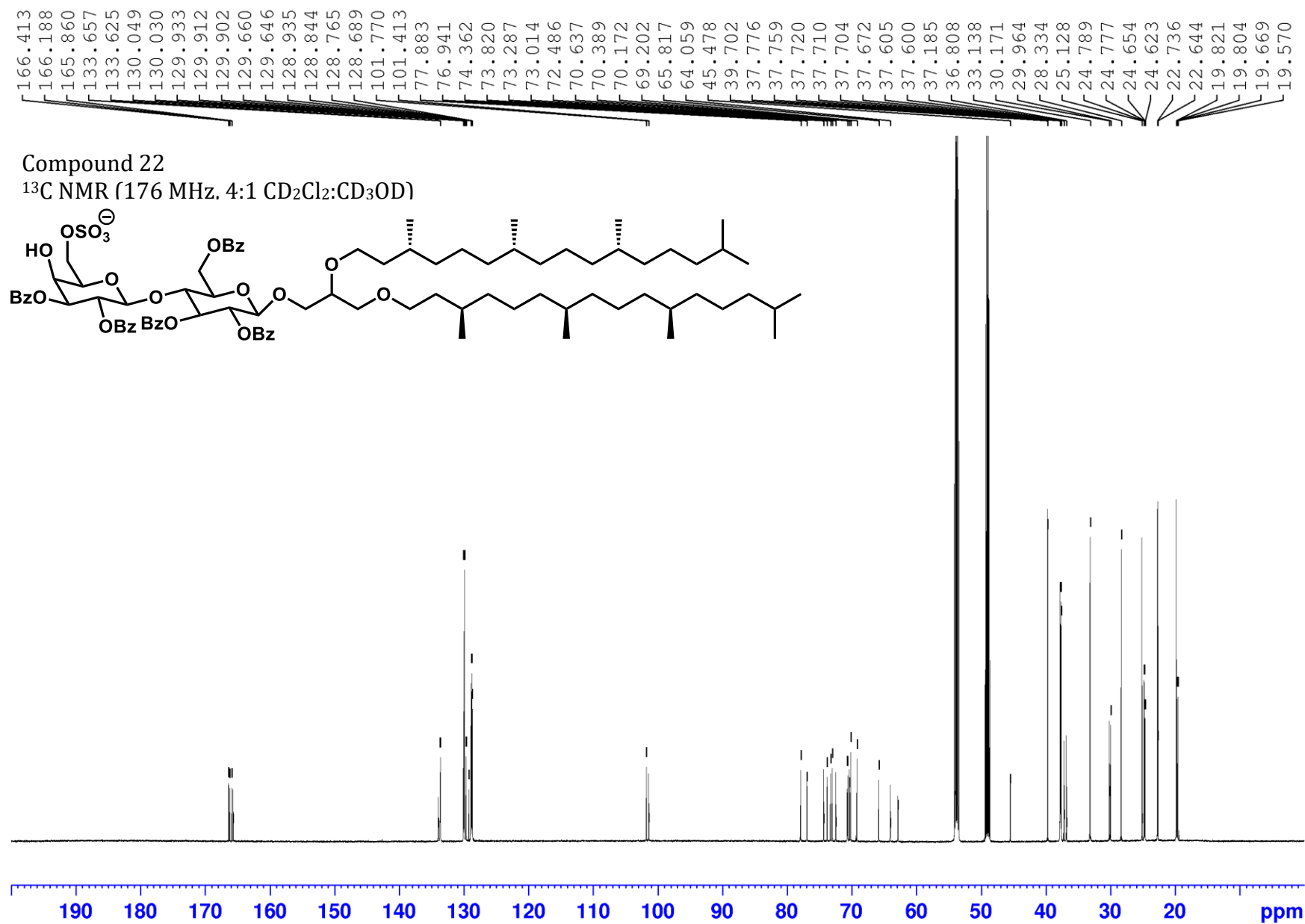




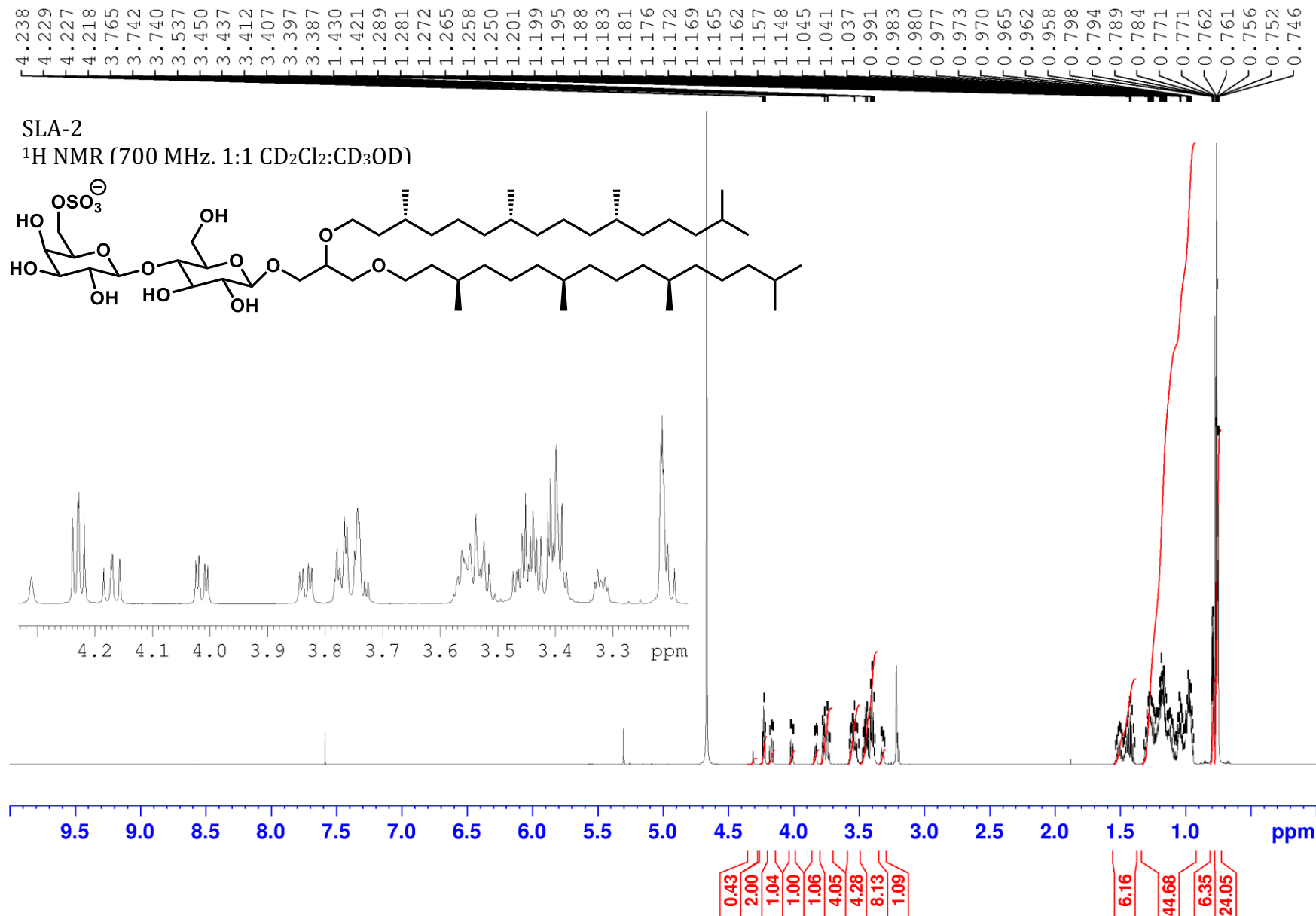
Compound 22

$^1\text{H}$  NMR (700 MHz, 4:1  $\text{CD}_2\text{Cl}_2:\text{CD}_3\text{OD}$ )

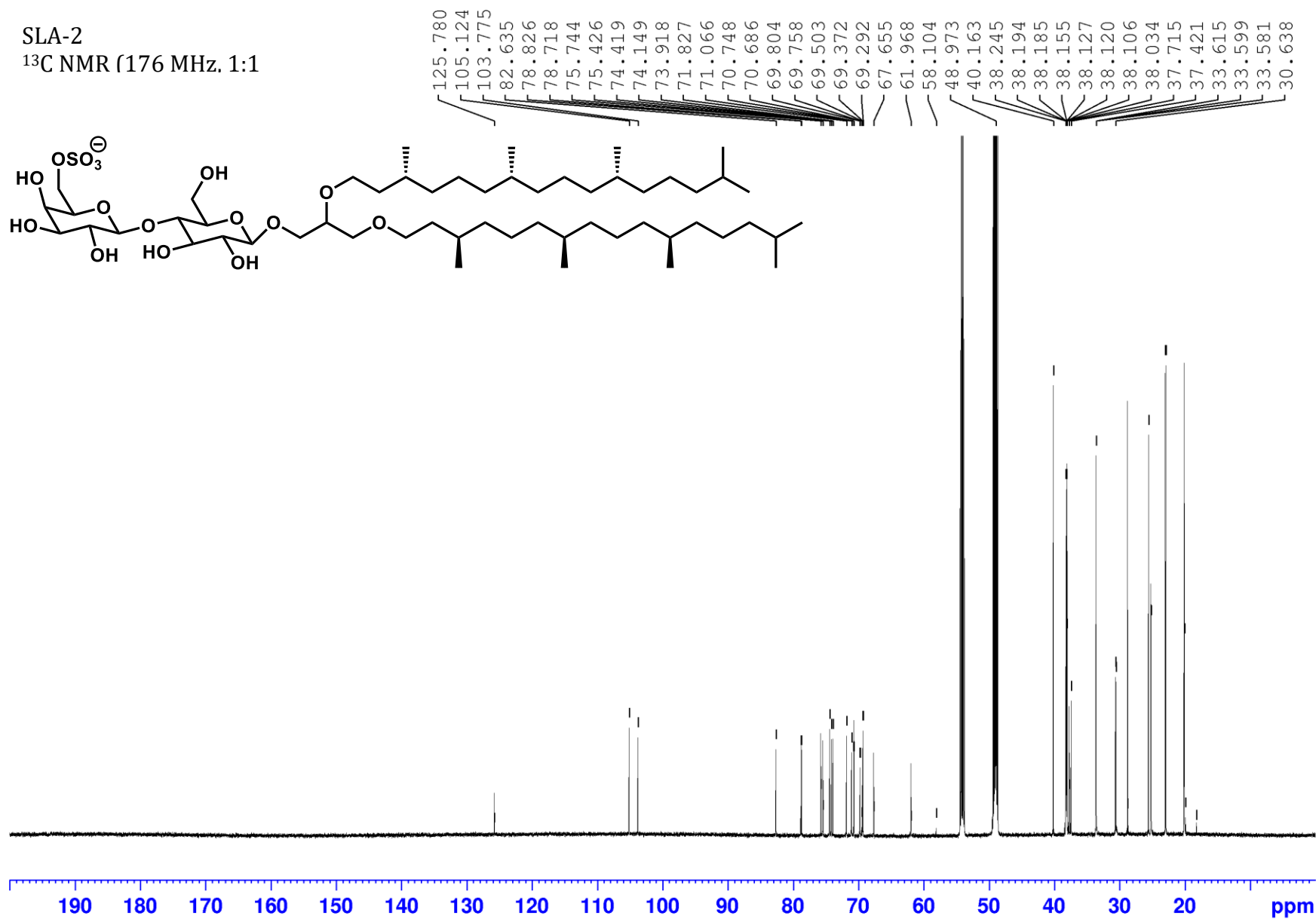


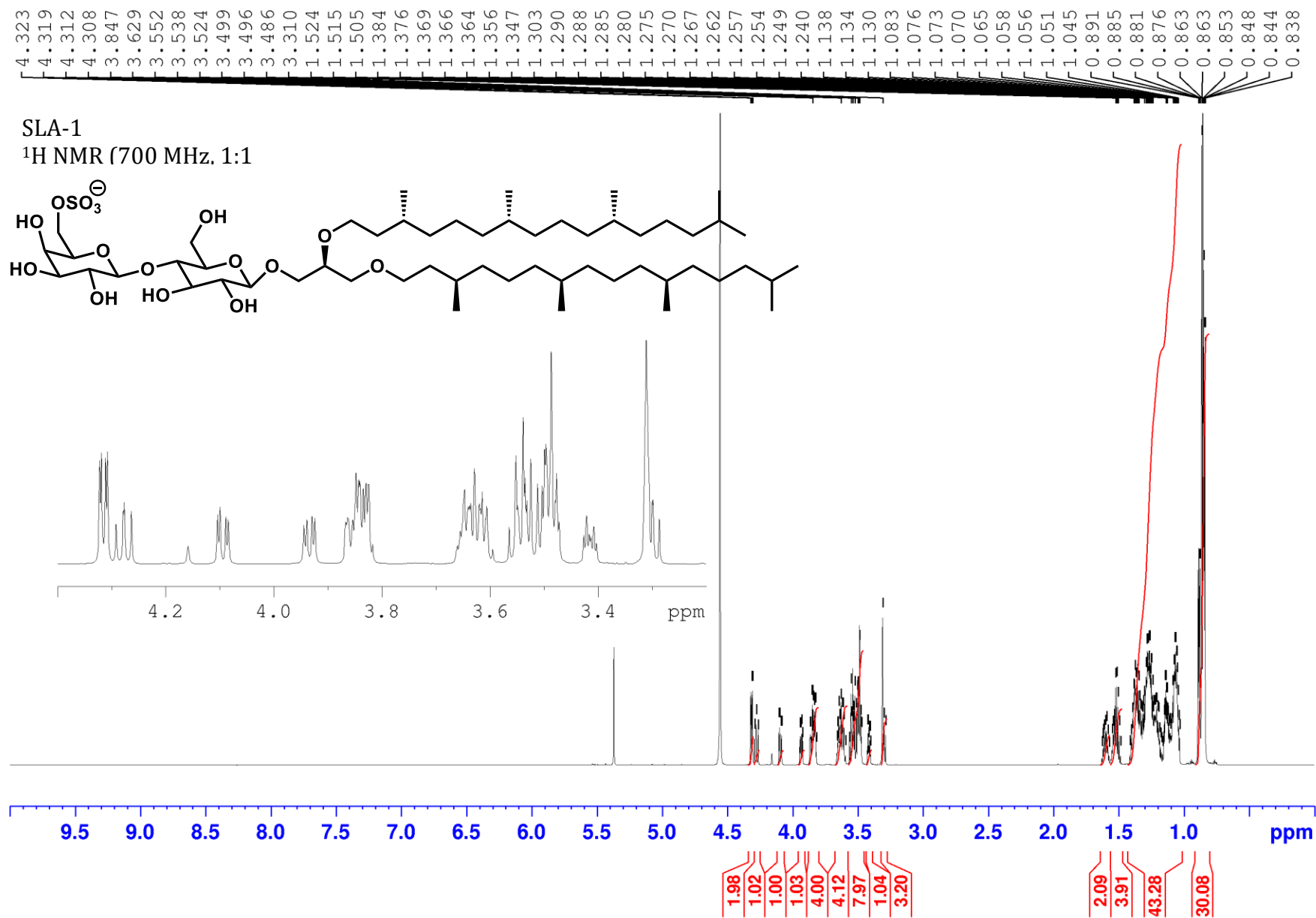


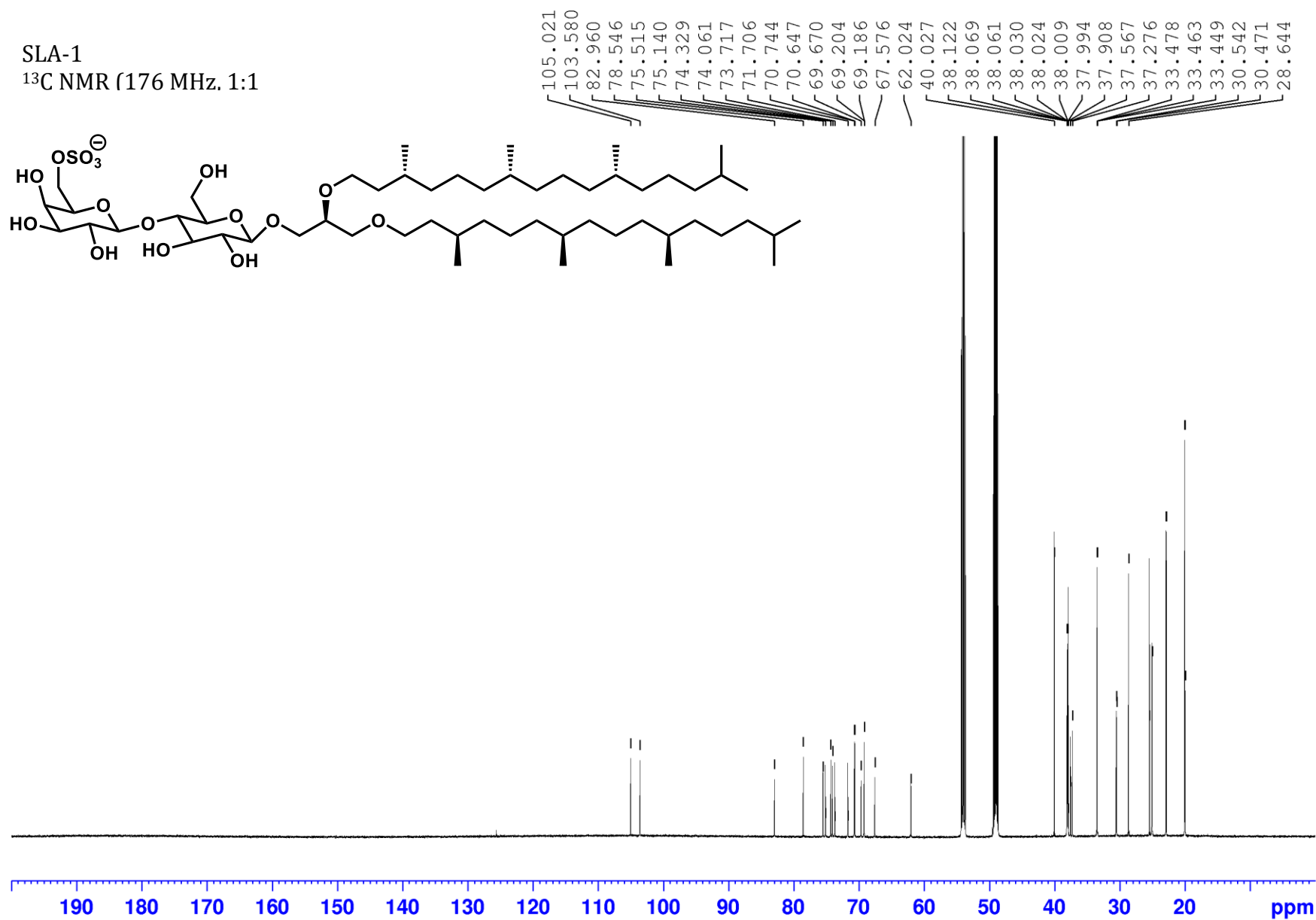




SLA-2  
 $^{13}\text{C}$  NMR (176 MHz, 1:1)





SLA-1  
<sup>13</sup>C NMR (176 MHz, 1:1)

## References

1. Whitfield, D. M.; Yu, S. H.; Dicaire, C. J.; Sprott, G. D., Development of New Glycosylation Methodologies for the Synthesis of Archaeal-Derived Glycolipid Adjuvants. *Carbohydrate Research* **2010**, *345* (2), 214-229.
2. Whitfield, D. M. S., G. D.; Krishnan, L. Sulfated-Glycolipids as Adjuvants for Vaccines. **2016**.
3. Prabha, S. P.; Karthik, C.; Chandrika, S. H., Phytol – a Biosurfactant from the Aquatic Weed Hydrilla Verticillata. *Biocatalysis and Agricultural Biotechnology* **2019**, *17*, 736-742.
4. Thakor, P.; Mehta, J.B.; Patel, R. R.; Patel, D. D.; Subramaniana, R. B.; Thakkar, V.R., Extraction and purification of phytol from Abutilon indicum: cytotoxic and apoptotic activity. *Royal Society of Chemistry Advances* **2016** (6) 48336–48345.